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(54) Title: NAPHTHALENE AMIDES HAVING LEUKOTRIENE-ANTAGONISTIC ACTION

$$R^{\frac{3}{4}} = \frac{1}{8} \frac{1}{6} \frac{1}{(CH_2)} - A - (CH_2) \frac{1}{n}$$

$$(CH_2) = B$$
(1)

(57) Abstract

Naphthalene amides of formula (I) wherein the substituent containing A is bound to the 6- or 7- position of the 2-naphthol system; the substinent containing B is bound to the benzene ring at any free position; R1 is hydrogen or methyl; R2 is hydrogen, fluorine, chlorine or -OCHs, which is bound to the naphthalene system at any positions except the 2- and the one occupied by the other substituent; R³ is hydrogen, fluorine, chlorine or bromine; A- is -CO-NR⁴- or -NR⁴-CO- group, wherein R⁴ is hydrogen or methyl; B is a 5-tetrazolyl or -COOR⁵ group, wherein R⁵ is hydrogen, a (C₁-C₄)-alkyl or a phenylalkyl group of less than 10 carbon atoms; m is 0 or 1; n and p are integers from 0 to 6, with the proviso that n + p is less or equal to 6; as well as the solvates and pharmaceutically acceptable salts thereof, have leukotriene antagonistic action.

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NAPHTHALENE AMIDES HAVING LEUKOTRIENE-ANTAGONISTIC ACTION

The present invention relates to novel naphthalene amides, the pharmaceutically acceptable salts and solvates thereof and pharmaceutical compositions containing them, having a leukotriene-antagonistic activity. The present invention also relates to a process for the preparation of the novel naphthalene amides, as well as to the therapeutic use thereof.

TECHNOLOGICAL BACKGROUND

It is well known that most eicosanoids, prostaglandins, leukotrienes and related compounds derive from 10 a fatty acid having 20 carbons and 4 unsaturations, called arachidonic acid (AA), which fundamentally esterifies the hydroxyl at the 2- position of the glycerol of the phospholipids contained in the cell 15 membranes. AA is released from the phospholipid containing it by the action of a lipase, phospholipase A2 (PLA2) ("CRC Handbook of Bicosanoids and Related Lipids", vol. II, Ed. A.L.Willis, CRS Press Florida (1989)). After being released AA is metabolized 20 in mammals mainly by two different pathways or enzyme systems. Through cyclooxygenase it produces prostaglandins and thromboxanes, the most significant being PGE2 and TxA2, which are directly involved in inflammation (Higgs et al. Annals of Clinical Research, 16, 287 (1984)). Through lipo-oxygenase it produces leuko-25 trienes, the most important being LTB4, and the peptideleukotrienes LTC4, LTD4 and LTB4. All of them are also involved inflammatory reactions, exhibiting in

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chemotactic activities, stimulating the secretion of lysosomic enzymes and playing an important role immediate hypersensitivity reactions (Bailey and Casey, Ann. Rep. Hed. Chem., 17, 203 (1982)). Leukotriene LTB, strong chemotactic agent which promotes the infiltration of leukocytes their subsequent and degranulation. (Salmon et al., Prog. Drug Res., 37, 9 (1991)). It has been widely shown that LTC, and LTD, have strong constrictive action on human bronchi (Dahlen 484 Nature, 288, (1980)), causing al., obstruction of airways by inflammation and mucus production (Marom et al., Am. Rev. Resp. Dis., 126, 449 (1982)), being thus involved in the pathogenesis of bronchial asthma, chronic bronchitis, allergic rhinitis. etc. Peptide-leukotrienes also bring about a blood extravasation caused by the increase of vascular permeability (Camp et al., Br. J. Pharmacol., 80, 497 (1883)) and are involved in some inflammatory diseases such as atopic eczema and psoriasis. On the other hand, several effects of peptide-leukotrienes on human cardiovascular system have been observed; they are mainly involved in the pathogenesis of the ischaemic cardiopathy. This relationship has been confirmed by the fact that coronary arteries can produce these mediators (Piomelli et al., J. Clin. Res., 33, 521A (1985)). These effects, together with the strong contractions observed in heart tissue caused by LTC, and LTD, suggest that these might contribute to other cardiovascular mediators disorders, such as coronary spasm, heart anaphylaxis, cerebral oedema and endotoxic shock.

From what said above it follows that the control of

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the biological activity of leukotrienes through compounds which inhibit their release or antagonize their effects, represents a new rational approach to the prevention, elimination or improvement of different allergic, anaphylactic, inflammatory and thrombotic conditions, in which such mediators are involved.

In literature some compounds have been described that can be considered as structurally related to the compounds of the present invention, having an inhibiting action on enzyme 5-lypoxygenase, and a leukotriene antagonistic activity. Kreft A.F. et al. described 2-[6-(2-quinolinylmethoxy)-2-naphthyl]propionic acid (US 4,960,892) and derivatives thereof (US 5,084,575), among which there are compounds containing sulfonimide, hydroxamic acid or hydroxamate groups, but the amides of the present invention are not included in their general formulas.

On the other hand, Stevenson D. et al. (US 4,579,866) described amides having an inhibiting action on 5-lypoxygenase, but differing from the compounds of invention in two aspects: the present first. contain a phenyl instead of a naphthyl; second, they contain an alkyl chain with only one carbon atom with or without branches between the amide function and the terminal carboxylic acid. In short, they are compounds containing a phenylalanine or glycine terminal residue.

The obtaining of compounds with high leukotriene antagonistic activity is still a problem in the therapy. The present invention provides a number of novel compounds that exhibit the above mentioned antagonistic action and that are useful in therapy.

DISCLOSURE OF THE INVENTION

The present invention relates to novel naphthalene amides of general formula I,

10 wherein:

the substituent containing A is bound to the 6- or 7position of the 2-naphthol system;

the substituent containing B is bound to the benzene ring at any free position;

15 -R¹ is hydrogen or methyl;

 $-R^2$ is hydrogen, fluorine, chlorine or $-OCH_3$, which is bound to the naphthalene system at any position except the 2- and the one occupied by the other substituent;

-R³ is hydrogen, fluorine, chlorine or bromine;

20 -A- is a -CO-NR⁴- or -NR⁴-CO- group, wherein R⁴ is hydrogen or methyl;

-B is a 5-tetrazolyl or $-COOR^5$ group, wherein R^5 is hydrogen, a (C_1-C_4) -alkyl or a phenylalkyl group of less than 10 carbon atoms;

25 m is 0 or 1;

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n and p are integers from 0 to 6, with the proviso that n + p is less or equal to 6.

The present invention also relates to a process for the preparation of the novel naphthalene amides, as well as the therapeutical use thereof.

The present invention also relates to the solvates

and the pharmaceutically acceptable salts of the amides of formula I and particularly the salts represented by formula Ia,

$$R^{\frac{3}{8}}$$

$$R^{\frac{1}{2}}$$

wherein M⁺ is an alkali metal cation (e.g. Na⁺, K⁺) or represents a half amount of an alkaline-earth metal cation (e.g. 1/2 Ca²⁺, 1/2 Mg²⁺), or a cation deriving from an amine or quaternary ammonium salt (such as triethanolammonium, tris(hydroxymethyl)methylammonium).

The compounds of formula I can have one or more asymmetric carbons in their structure. The present invention comprises all the possible stereoisomers as well as the mixtures thereof.

Preferred compounds are those wherein \mathbb{R}^2 is hydrogen and B is a 5-tetrazolyl or \mathbb{C}^5 group, wherein \mathbb{R}^5 is hydrogen, methyl, ethyl or benzyl.

Also preferred are the compounds of formula I wherein \mathbb{R}^3 is hydrogen or chlorine and -A- is -CONH- or -NHCO-.

When the substituent containing A is bound to the 6- position of the 2-naphthol system, particularly preferred are the compounds of formula I wherein R¹ is hydrogen, m is 1, and -A- is -NHCO-; or those wherein -A- is -CONH-, being n and p integers ranging between 0 and 3.

When the substituent containing a A is bound to the

7- position of the 2-naphthol system, particularly preferred are the compounds of formula I wherein \mathbb{R}^1 is hydrogen, m is 1 and -A- is -CONH-; or also those wherein m is 0, -A- is -CONH-, being n and p integers

- ranging between 0 and 3.

 Particularly preferred compounds of the present invention are the following ones: N-[4-(1H-5-tetra-zoly1)phenylmethy1]-2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamide;
- methoxy)-2-naphthyl]propanamide;

 N-[4-(1H-5-tetrazolyl)methylphenyl]-2-[6-(2-quinolinyl-methoxy)-2-naphthyl]propanamide (sodium salt);

 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]-benzoic acid;
- 4-[4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]phenyl]butanoic acid;
 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyl]benzoic acid;
- 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyl]benzoic acid;
- 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]phenylacetic acid;
 - 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]-phenylacetic acid;
- 30 4-[4-[2-[6-[(7-chloro-2-quinolinyl)methoxy]-2-naphthyl]propanamido]phenyl]butanoic acid;

N-[4-(1H-5-tetrazolyl)phenylmethyl]-6-(2-quinolinylme-thoxy)-2-naphthaleneacetamide;

4-[4-[[6-(2-quinolinylmethoxy)-2-naphthyl]carboxamido]-phenyl]butanoic acid;

5 N-[3-(1H-5-tetrazolyl)phenylmethyl]-6-(2-quinolinylme-thoxy)-2-naphthalenecarboxamide;

4-[4-[[7-(2-quinolinylmethoxy)-2-naphthyl]carboxamido]-phenyl]butanoic acid;

N-[4-(1H-5-tetrazolyl)phenylmethyl]-7-(2-quinolinylme-

10 thoxy)-2-naphthalenecarboxamide;

4-[2-[[7-(2-quinolinylmethoxy)-2-naphthyl]carboxamido]-ethyl]benzoic acid;

N-[4-(1H-5-tetrazolyl)phenylethyl]-7-(2-quinolinylme-thoxy)-2-naphthalenecarboxamide;

4-[4-[[6-(2-quinolinylmethoxy)-2-naphthyl]methylaminocarbonyl]phenyl]butanoic acid;

N-[4-(1H-5-tetrazolyl)phenylpropyl]-7-(2-quinolinylme-thoxy)-2-naphthalenecarboxamide;

as well as the carboxylic acid esters described in the examples.

According to the present invention, the compounds of general formula I wherein A is -CO-NR⁴- are obtained by a process in which, starting from a compound of general formula II,

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$$\mathbb{R}^{\frac{3}{2}} \mathbb{R}^{\frac{1}{2}}$$

$$\mathbb{R}^{\frac{1}{2}} \mathbb{R}^{\frac{1}{2}}$$

$$\mathbb{R}^{\frac{3}{2}} \mathbb{R}^{\frac{1}{2}} \mathbb{R}^{\frac{1}{2}}$$

$$\mathbb{R}^{\frac{3}{2}} \mathbb{R}^{\frac{1}{2}} \mathbb{R}^{\frac{1}{2}}$$

$$\mathbb{R}^{\frac{3}{2}} \mathbb{R}^{\frac{1}{2}} \mathbb{R}^{\frac{1}{2}}$$

30 wherein R^1 , R^2 , R^3 and m have the above defined meanings, is reacted with a compound III,

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5 III

wherein R4, n and p have the above defined meanings and D can be equivalent to the group B in I or, when B in formula I is COOH, then D contains a suitable carboxyprotecting group, for example as a methyl, ethyl or benzyl ester. The reaction between II and III is carried out in the presence of a carboxy-activating agent such dicyclohexylcarbodiimide 25 or methylamino)propyl]-3-ethylcarbodiimide and a base such triethylamine or 4-dimethylaminopyridine, suitable aprotic solvent such as chloroform, methylene chloride or N,N-dimethylformamide, at a temperature ranging between 0° and 40°C for a time from 3 to 24 hours. In this way, a compound of formula IVa is obtained.

$$\begin{array}{c|c}
R^{2} & R^{1} & O \\
\downarrow & \downarrow & O \\
R^{3} & \downarrow & O \\
R^{3} & \downarrow & O \\
R^{4} & \downarrow & O \\
\end{array}$$

$$\begin{array}{c|c}
R^{2} & O \\
\downarrow &$$

IVa

this compound coincides with I or is converted into I removing any COOH-protecting groups present in D, then, when D is for example a methyl, ethyl or benzyl ester, can be removed by treatment with a suitable base such as lithium or sodium hydroxide in aqueous solution, in a suitable organic solvent such as methanol, ethanol or

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tetrahydrofuran, at a temperature ranging between 20°C and the solvent reflux, for a time from 1 to 48 hours.

A compound of general formula I wherein A is $-NR^4$ -CO- is obtained according to the same process as above, starting from the compounds V and VI,

HOOC-(CH₂) n CH₂ PD

wherein R^1 , R^2 , R^3 , R^4 , D, m, n and p have the above defined meanings. In this way a compound of formula IVb,

which coincides to I or is converted into I removing any COOH-protecting groups present in D, is obtained as described above.

When a given salt of general formula Ia is desired, a compound I can be treated with a suitable base or ion-exchanger, according to the conventional chemical techniques. Thus, for example, I can be treated with sodium hydroxide or tris(hydroxymethyl)methylamine in a suitable solvent such as water-methanol or ethanol mixtures, for a time from 15 min to 2 hours, at a

temperature ranging between 25°C and the solvent reflux.

A starting product of formula II can be obtained, for example, following the synthesis sequence shown in scheme 1.

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$$R^{3} = \frac{R^{2}}{(CH)_{m}} - COOMe \qquad (2)$$

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$$R^{\frac{3}{4}}$$
 $R^{\frac{1}{2}}$
 $R^{\frac{1}{2}}$

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In this synthetic sequence, a compound IX can be obtained, for example, subjecting a compound VII to the action of a base such as sodium methoxide or sodium hydride, thereafter reacting it with a compound VIII, wherein R³ represents the groups defined above and X is a bromine or chlorine atom, in a suitable organic solvent such as benzene, N,N-dimethylformamide or tetrahydrofuran, at a temperature ranging between 0° and 25°C for a time from 3 to 24 hours (step (1)).

A compound II can be obtained starting from IX (step (2)) by basic hydrolysis as described for the preparation of I with B=COOH starting from IVa.

Analogously, a starting compound of general formula V can be obtained, for example, following the synthetic sequence shown in scheme 2.

Scheme 2

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$$R^{\frac{3}{4}} \stackrel{\text{N}}{\longrightarrow} 0$$

$$R^{\frac{1}{2}} \stackrel{\text{R}^{1}}{\longrightarrow} -N - Ac$$

$$XI$$

$$(4)$$

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$$\begin{array}{c|c}
R^2 & R^1 \\
\hline
(CH)_{m} & NH \\
\hline
R^4
\end{array}$$

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synthetic sequence a compound this obtained starting from compound X (step (3)), a following the same described for the process 8.5 preparation of IX.

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A compound V can be obtained starting from XI by hydrolysis with hydrochloric or sulfuric acid, in a suitable solvent such as tetrahydrofuran or dioxane, at a temperature ranging between 25° and the solvent reflux, for a time from 1 to 24 hours (step (4)).

30 scheme 3.

A starting compound VII can be obtained, for example, following the synthetic sequence shown in

Scheme 3

VIIc (VII with $R^1=CH_3$ and m=1)

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In this sequence the starting product is a compound of general formula XII, wherein R2 represents the groups described above and P represents a suitable hydroxymethyl or such as group, protecting bе Said compound t-butyldimethylsilyl group. corresponding easily starting from the obtained hydroxynaphthoic acids, following preparation processes described in literature (Gray G.W. et al., J. Chem. Soc., 678 10 (1954) and Daines et al., WO 9217172).

Starting from XII, VIIa (VII with m=0) can be obtained removing the protecting group P (step (5)); for example when this is a methyl group, it can be removed by means of boron tribromide in an organic solvent such as methylene chloride or chloroform, at a temperature ranging between -78° and 0°C for a time from 2 to 8 hours. When P is a t-butyldimethylsilyl group, it can be removed with a base such as potassium carbonate or bicarbonate, in a solvent such as tetrahydrofuran, methanol or dioxane at a temperature ranging between 0° and 50°C, for a time from 2 to 12 hours.

A compound VIIb (VII with R¹=H and m=1) can be obtained also starting from a compound XII. By means of a step (6), a compound XII is subjected to the action of a suitable metal hydride, such as sodium borohydride, in an solvent such as methanol, ethanol or tetrahydrofuran, in the presence of a catalytic amount of water, at a temperature ranging between 20°C and the solvent reflux, for a time from 3 to 24 hours: in this way compound XIII is obtained. This compound is subjected to a tosylation reaction (step (7)) with tosyl chloride in the presence of pyridine or triethylamine in methylene chloride, to

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give a compound of formula XIV, which is treated with sodium cyanide in tetrahydrofuran or DMSO at a temperature between 25°C and the solvent reflux to yield compound XV (step (8)). The hydrolysis of XV, for example with NaOH in ethanol at a temperature ranging between 25°C and the solvent reflux, followed by the esterification with methanol in the presence of sulfuric acid, gives compound XVII (steps (9) and (10) respectively). Finally, the removal of the protecting group P, as described above, can lead to a compound VIIb.

Analogously, a compound VIIc (VII with R¹=CH₃ and m=1) can be obtained by a synthesis sequence in which the starting product is a compound XV. By treating XV with a strong base, such as sodium methoxide or potassium t-butoxide, in a suitable solvent such as tetrahydrofuran or N,N-dimethylformamide, at a temperature ranging between -30° and 25°C for a time from 2 to 24 hours (step (12)), a compound XVIII is obtained. After that, by hydrolysis of the nitrile group present in XVIII (step (13)), esterification of the carboxylic group of XIX (step (14)) and elimination of the protecting group P in XX (step (15)), as described above, compound VIIc is obtained.

25 Alternatively, a compound VIIb can be obtained following the synthesis shown in scheme 4.

Scheme 4

XXI

IIXX

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VIIb (VII with R1=H and m=1)

Starting from a compound of general formula XXI, which is prepared easily following processes described in literature (Müller et al. Helv. Chim. Acta, 57, 790 (1974)), a compound XXII can be prepared by the modified Willgerodt's reaction. By means of such a reaction, treating a compound XXI with sulfur in the presence of an amine such as morpholine, which in its turn acts as the solvent, at the solvent reflux temperature, for a time from 6 to 24 hours, a compound of formula XXII (step (16)) is obtained after a suitable treatment with hydrochloric acid. The removal of the hydroxy-protecting methyl group, according to the process described above, leads to the preparation of a compound VIIb (step (17)).

Alternatively, a compound VIId, equivalent to VII with R^1 =CH₃, R^2 =H, m=1 and with an alkylmethoxycarbonyl substituent at the 6- position of the 2-naphthol system can be prepared, starting from a compound XXIII, which is the known NSAID naproxen, commercially available both as the racemate and in the form of the two resolved enantiomers. Compound XXIII is subjected to demethylation with BBr₃ in a solvent such as chloroform

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or methylene chloride at a temperature between -78°C and room temperature for 1 to 6 hours, followed by addition of methanol to the reaction, to obtain the desired compound VIId (scheme 5).

Scheme 5

VIId (VII with R¹=CH₃, R²=H, m=1 and a methoxycar-bonylalkyl substituent at the 6-position)

Alternatively a compound IIa, i.e. a compound II with m=0, can be obtained following the process described in scheme 6.

Scheme 6

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$$R^{2} \longrightarrow CN$$

$$VIII$$

$$R^{3} \longrightarrow N$$

$$XXIV$$

$$XXIV$$

$$R^{2} \longrightarrow CN$$

$$XXV$$

$$R^{2} \longrightarrow COOH$$

IIa (II with m=0)

Starting from a compound XXIV, easily available following chemical processes described in literature (Tolbert et al. J. Am. Chem. Soc., 112, 8163 (1990)), a

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compound XXV can be prepared by condensation with VIII (step (18)) following the method described for the preparation of the compounds of general formula IX. The hydrolysis of XXV (step (19)), according to the process described for the preparation of XIX, leads to the compounds of general formula IIa.

A compound of general formula Xa, i.e. of general formula X with $R^1=H$, $R^4=H$ and m=1, can be obtained, for example, following the synthetic sequence shown in scheme 7.

Scheme 7

A compound XXVII can be obtained starting from a compound of general formula XXVI, commercial or easily available starting from XXIV, by catalytic hydrogenation with a suitable catalyst, such as Pd-C or Pd(OH)₂-C in a suitable solvent such as ethanol or methanol, in the presence of an acid such as acetic acid or hydrochloric

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acid, under hydrogen pressures ranging between atmosphere pressure and 2 kg/cm^2 , at a temperature from 25° to 50°C, for a time from 3 to 24 hours (step (20)).

A compound XXVIII can be obtained by acetylation of XXVII for example with acetic anhydride in the presence of a base such as pyridine or triethylamine in a solvent such as chloroform or methylene chloride, at a temperature between -78° and 25°C for a time from 1 to 6 hours (step (21)).

A compound Xa can be obtained starting from XXVIII by deprotecting the methylated hydroxy group (step (22)) according to the process described above. Analogously, a compound Xb, i.e. a compound of general formula X with R1=H,, R4=CH3 and m=1, can be obtained by methylation of a compound XXVIII (step (23)) with methyl iodide or methyl sulfate in the presence of a base such as lithium disopropylamide, lithium amide or potassium t-butoxide in a suitable solvent such as tetrahydrofuran, benzene or toluene at a temperature ranging between -78° and 25°C for 2-6 hours. The subsequent deprotection of the methylated hydroxy group leads to a compound Xb (step (24)).

A compound Xc, i.e. a compound of general formula X with $R^1=CH_3$ and m=1, can be obtained for example following the synthetic sequence shown in scheme 8.

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The reduction of a compound of formula XXX (easily available starting from XXI), for example, with a suitable hydride such as sodium borohydride in a solvent such as methanol, ethanol or tetrahydrofuran at a temperature between 20° and the solvent reflux for a time from 3 to 24 hours, gives a compound XXXI (step (25)).

XXXIV

Xc (X with R'=CH2 and m=1)

A compound XXXII can be obtained, for example, by treatment of a compound XXXI (step (26)) with p-toluenesulfonic acid chloride in the presence of a base such as pyridine or triethylamine which can in their turn act as solvents, the presence of other solvents such as chloroform or methylene chloride being optional, at a temperature ranging between 20° and 40°C for a time from 3 to 18 hours.

A compound of formula XXXIII can be prepared for example starting from a compound XXXII (step (27)) by

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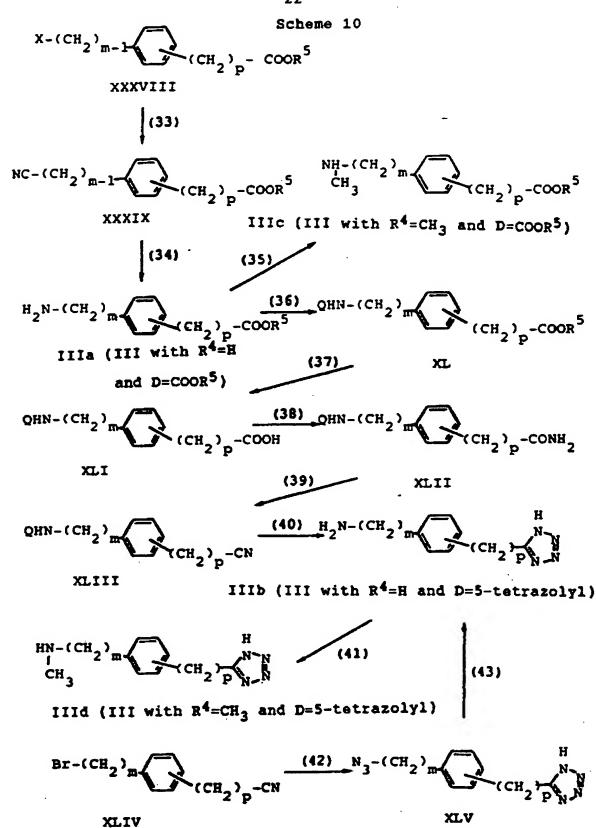
replacing the tosylate group with an azido group, using sodium azide as the reactive, in a suitable organic solvent such as N,N-dimethylformamide or tetrahydrofuran at a temperature ranging between 20°C and the solvent reflux for a time from 6 to 24 hours; and following catalytic hydrogenation of the intermediate azide with a suitable catalyst such as Pd-C or in a solvent such as ethanol, methanol or ethyl acetate, under hydrogen pressures ranging between atmosphere pressure and 2 kg/cm², at a temperature between 0° and 50°C, for a time from 6 to 24 hours.

Finally a compound Xc can be obtained by successive steps of acetylation, optional methylation of the amido group and deprotection of the methylated hydroxyl, as described above for the compounds Xa and Xb.

Analogously, a compound Xd, i.e. of general formula X with m=0, can be prepared starting from XXXV, easily available according to processes described in literature (Airan et al., J. Am. Chem. Soc., 26, 339 (1927)), by the process shown in scheme 9, in which a synthesis sequence similar to that described for the preparation of the other compounds Xa-c is used.

Scheme 9

A starting compound III can be prepared following the synthesis sequence shown in scheme 10.



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starting from a compound XXXVIII, commercial or easily available according to similar chemical processes, wherein m and p have the values described above, R⁵ represents the groups described above, except hydrogen and X can be a halogen atom, a compound XXXIX can be prepared by treatment with NaCN in a suitable solvent such as dimethylsulfoxide, water or ethanol, at a temperature ranging between 25°C and the solvent reflux for 1-8 hours (step 33). By reduction of the cyano group in XXXIX, following the same process as described in step (20), a compound IIIa ca be prepared, which is equivalent to a compound of general formula III with R⁴=H and D=COOR⁵.

Starting from a compound IIIa, a compound IIIb can be obtained, i.e. of general formula III with R4=H and D=5-tetrazolyl, by a process which comprises the steps (36)-(40). First, the amino group in IIIa is protected with a benzyloxycarbonyl or t-butoxycarbonyl group, according to processes widely described in literature, to yield compound XL, wherein Q represents one of the amino-protecting groups mentioned above. The basic hydrolysis of XL leads to the preparation of XLI, starting from which a compound XLII can be prepared by example, of a reaction with means, for chloroformate in the presence of a base such triethylamine or pyridine in a solvent tetrahydrofuran or ethyl ether and following treatment with ammonia, at a temperature between 0° and 25°C, for a time from 30 min to 3 hours. The dehydration of a compound XLII, for example with phosphorous oxychloride, in a solvent such as N,N-dimethylformamide at a

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temperature ranging between 0° and 50°C for 3-24 hours, gives a compound XLIII. The treatment of a compound XLIII with sodium azide in a suitable solvent such as N.N-dimethylformamide at a temperature ranging between the reflux and solvent 25°C and the elimination of the protective group present in the amino group, according to conventional techniques, yields a compound IIIb. Alternatively, when a compound XLIV is commercial or easily available by similar synthetic methods, a compound IIIb can be obtained starting from XLIV by reaction with sodium azide (step (42)) and subsequent reduction of the azide (step 43), described above in step (27).

Starting from a compound of formula IIIa or of formula IIIb, a compound IIIc or IIId can be obtained respectively by methylation of the amine according to, for example, a process comprising first the formylation amino group, with acetic anhydride of the formaldehyde mixtures in a suitable solvent, such as tetrahydrofuran or ethyl ether, at a temperature between 0° and 25°C for 3-24 h, followed by reduction of the formyl group with the BH2-tetrahydrofuran complex in a solvent such as tetrahydrofuran or ethyl ether at a temperature ranging between -78° and 0°C for 6-24 hours.

A starting compound of formula VI can be prepared for example following the process shown in scheme 11.

Starting from a compound XLVI, wherein m and p have the meanings described above and R⁵ also represents the groups described above, except hydrogen, and Y represents a chlorine or bromine atom, commercially available in some instances or easily prepared by

similar synthetic methods, VIa, i.e. the compound of general formula VI with D=COOR⁵, (steps (44) and (45)), can be obtained following the processes described above for steps (33) and (13).

20. VIb (VI with D=5-tetrazoly1)

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Starting from XLVI, wherein Y represents a hydroxyl group, a compound XLVIII can be prepared following a process similar to that described in scheme 10. Finally, a compound VIb, i.e. of general formula VI with D=5-tetrazolyl, can be prepared by a process which comprises the tosylation of XLVIII, the substitution of the tosylate by a nitrile group and the final hydrolysis according to the above described methods.

Alternatively, when in VI m=0, D=COOR⁵ and the two
substituents of the benzene ring are in para position,
compopund of formula VIc can be prepared following the

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synthesis shown in scheme 12.

Scheme 12

$$(CH_2)_p$$
- $COOR^5$ (47) OHC $(CH_2)_p$ - $COOR^5$

XLIX

 (48)
 $(CH_2)_p$ - $COOR^5$

VIc (VI with m=0, D=COOR⁵ and para orientation of the substituents)

A compound L can be prepared starting from XLIX, for example, by a formylation reaction using a suitable reactive such as hexamethylenetetramine or N,N-dimethylformamide in the presence of trifluoroacetic acid or phosphorous oxychloride at a temperature ranging between 25° and 100°C, for a time from 2 to 24 hours. The subsequent oxidation with a suitable oxidizing agent such as Jones's reagent at a temperature ranging between 0° and 25°C for a time of 2 to 18 hours, leads to the desired compound VIc.

The compounds of the present invention show a marked antagonistic activity of leukotrienes effects and they have therefore anti-inflammatory and anti-allergic properties which make them useful in the treatment of diseases wherein those mediators are involved.

Said compounds can be therefore used in human therapy, for the prevention and treatment of allergic rhinitis, bronchial asthma, hypersensitivity reactions such as allergic conjunctivitis, various inflammatory conditions such as rheumatoid arthritis, osteoarthritis,

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tendinitis, bursitis, psoriasis and related inflammations.

The compound of the present invention may also be used in the treatment of diseases of the cardiovascular system, such as cardiac ischemia, myocardic infarct, coronary spasm, cardiac anaphylaxis, cerebral oedema and endotoxic schock.

For the intended therapeutic uses, the compounds of the invention are formulated in suitable pharmaceutical compositions, using conventional techniques and methods, as disclosed in Remington's Pharmaceutical Science Handbook, Mack Pub. Co., N.Y. U.S.A. Examples of said formulations include capsules, tablets, syrups and the like, containing from 1 to 1000 mg of active principle per unit dose.

EXAMPLES

The following examples illustrate the preparation and the pharmacological activity of the compounds of the present invention.

20 Example 1: N-[4-(1H-5-tetrazolyl)phenylmethyl]-2[6-(2-quinolinylmethoxy)-2-naphthyllpropanamide

1A Methyl 2-(6-hydroxy-2-naphthyl)propionate

Boron tribromide (70 ml, 737 mmol) was added at -78°C to a solution of 2-(6-methoxy-2-naphthyl)propionic acid (10 g, 43.4 mmol) in dry methylene chloride (130 ml). The reaction mixture was stirred at room temperature for 5 h, then 136 ml of methanol were added and stirred for 18 hours. After this time, the mixture was evaporated to dryness, water (250 ml) was added and extracted with ethyl ether (4x100 ml). The combined ether phases were dried and the solvent was evaporated

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off, to obtain a crude which was purified by flash chromatography through a silica gel column. Eluting with petroleum ether:ethyl ether, 9:1, 6.9 g of the title compound were obtained as a white solid with melting point 184-186°C (70% yield).

¹H N.M.R. (300 MHz, CDCl₃) & ppm: 1.56 (d, 3H); 3.70 (s, 3H); 3.87 (q, 1H); 5.75 (s, 1H); 7.08 (m, 2H); 7.38 (dd, 1H); 7.60 (d, 1H); 7.66 (m, 2H).

1B Methyl 2-[6-(2-quinolinylmethoxy)-2-naphthyllpropionate

Sodium methoxide (6.0 ml, 31.2 mmol) was added to a solution of methyl 2-(6-hydroxy-2-naphthyl)propionate (7.2 g, 31.2 mmol) in DMF (200 ml) and stirred at room temperature for 5 min. that 2-chloromethylquinoline (5.5 g, 31.2 mmol) was added 15 thereto and the reaction mixture was stirred at room temperature for 18 h, then evaporated to dryness, the residue was dissolved in ethyl acetate (250 ml), washed with 5% $NaHCO_3$ (3x25 ml), dried and the solvent was 20 evaporated off, to obtain a crude which was purified by crystallization with methanol. 9.3 g of the title compound wre obtained as a white solid with melting point 98-99°C (80% yield).

1H N.M.R. (300 MHz, DMSO) 6 ppm: 1.42 (d, 3H); 3.53 (s,
25 3H); 3.87 (q, 1H); 5.44 (s, 2H); 7.30 (m, 2H); 7.39 (d,
1H); 7.56 (t, 1H); 7.67 (m, 5H); 7.93 (d, 1H); 7.99 (d,
1H); 8.37 (d, 1H).

1C 2-[6-(2-Quinolinylmethoxy)-2-naphthyllpropionic acid

1M lithium hydroxide (31.7 ml) was added to a 30 solution of methyl 2-[6-(2-quinolinylmethoxy)-2-naphthyl]propionate (5 g, 13.2 mmol) in THF (40 ml) and

stirred at room temperature for 48 h. After that THF was evaporated off, pH was adjusted to 4-5 with 1M HCl and the mixture was extracted with ethyl acetate, to obtain 4.7 g of the title compound as a white_solid with melting point 192-194°C (99% yield).

¹H N.M.R. (300 MHz, DMSO) & ppm: 1.39 (d, 3H); 3.76 (q, 1H); 5.44 (s, 2H); 7.29 (dd, 1H); 7.35 (dd, 1H); 7.40 (d, 1H); 7.58 (t, 1H); 7.69 (m, 3H); 7.77 (dt, 1H); 7.80 (d, 1H); 7.95 (d, 1H); 7.98 (d, 1H); 8.38 (d, 1H).

10 <u>1D 4-(1H-5-Tetrazolyl)azidomethylbenzene</u>

Sodium azide (15 g, 230 mmol) and ammonium chloride (12.3 g, 230 mmol) were added to a solution of 4-bromomethylbenzonitrile (5 g, 25.5 mmol) in DMF (75 ml). The reaction mixture was stirred at 110°C for 16 h, then poured onto 200 ml of 1M HCl and extracted with ethyl acetate (4x75 ml). The organic phase was dried and solvents were removed, to obtain a oil which was diluted with ethyl ether and petroleum ether, to give a precipitate which was filtered and washed with petroleum ether. In this way 5 g of the title compound were obtained (97% yield).

¹H N.M.R. (300 MHz, CD₃OD) 6 ppm: 4.52 (s, 2H); 7.59 (d, 2H); 8.07 (d, 2H).

1E 4-(1H-5-Tetrazolyl)benzylamine hydrochloride

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1.4 g of 10% palladium on charcoal was added to a solution of 4-(1H-5-tetrazolyl)azidomethyl-benzene (4.3 g, 21.5 mmol) in methanol (400 ml) and concentrated HCl (14 ml) and stirred at room temperature for 4 days, under hydrogen atmosphere. After that the reaction mixture was filtered and the filtrate was evaporated to dryness to obtain a crude which was redissolved in hot

methanol, crystallizing in this way 3.7 g of the title compound as a yellowish solid with melting point >360°C (80% yield).

1_{H N.M.R.} (300 MHz, DMSO) & ppm: 4.16 (d, 2H); 7.75 (d, 2H); 8.18 (d, 2H); 8.56 (s, 3H).

1F N-[4-(1H-5-Tetrazolyl)phenylmethyl]-2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamide

4-(1H-5-tetrazolyl)benzylamine hydrochloride (0.212 g, 1.0 mmol), N,N-dimethylaminopyridine (0.373 g, 3.05 1-[3-(dimethylamino)propyl]-3-ethylcarbodi-10 mmol) and imide hydrochloride (0.202 g, 1.05 mmol) were added in solution of succession to 2-[6-(2-quinolinylmethoxy)-2-naphthyl]propionic acid (0.376 g, 1.05 mmol) in dry methylene chloride (50 ml). The reaction mixture was stirred at room temperature for 15 24 h, thereafter was poured onto water (30 ml) and pH was adjusted to 5 with 1M HCl, the phases were separated and the organic one was extracted with ethyl acetate (4x50 ml). The combined organic extracts were dried and the solvent was evaporated off, to obtain a crude which 20 crystallization in metha-Was purified by nol-chloroform-acetic acid mixtures, thereby obtaining 0.351 g of the title compound as a white solid with melting point 214-215°C (65% yield).

30 Example 2: N-[3-(1H-5-tetrazolyl)phenylmethyl]-2-

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[6-(2-quinolinylmethoxy]-2-naphthyl]propanamide

Following the process described in example 1 (point F), starting from 2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and 3-(1H-5-tetrazo-lyl)benzylamine hydrochloride, the title compound was prepared as a white solid with melting point 114-115°C (70% yield).

¹H N.M.R. (300 MHz, DMSO) δ ppm: 1.48 (d, 3H); 3.86 (q, 1H); 4.38 (m, 2H); 5.52 (s, 2H); 7.33 (dd, 1H); 7.47 (m,

10 4H); 7.66 (dt, 1H); 7.80 (m, 6H); 7.98 (s, 1H); 8.04 (d, 1H); 8.09 (d, 1H); 8.47 (d, 1H); 8.66 (t, 1H).

Example 3: N-[2-(1H-5-tetrazolyl)phenylmethyl]-2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamide

Following the process described in example 1 (point 15 F) starting from 2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and 2-(1H-5-tetrazo-lyl)benzylamine hydrochloride, the title compound was prepared as a yellowish solid with melting point 189-190°C (57% yield).

Example 4: N-(4-cyanomethylphenyl)-2-[6-(2-quinolinyl-methoxy)-2-naphthylpropanamide

Following the process described in example 1 (point F), starting from 2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and 4-cyanomethylaniline, the title compound was prepared as a white solid (recrystallized from methanol) with melting point 177-178°C (77% yield).

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1H N.M.R. (300 MHz, DMSO) & ppm: 1.45 (d, 3H); 3.90 (m, 3H); 5.43 (s, 2H); 7.20 (d, 2H); 7.27 (dd, 1H); 7.39 (d, 1H); 7.46 (d, 1H); 7.56 (m, 3H); 7.69 (t, 2H); 7.48 (m, 2H); 7.81 (d, 1H); 7.94 (d, 1H); 8.00 (d, 1H); 8.36 (d, 1H); 10.13 (s, 1H).

Example 5: N-[4-(1H-5-tetrazolyl)methylphenyl]-2-[6-(2-guinolinylmethoxy)-2-naphthyl]propanamide

Pollowing the process described in example 1 (point D), starting from N-(4-cyanomethylphe-nyl)-2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamide, the title compound was prepared as a white solid (recrystallized from methanol-ethyl acetate) with melting point 179.1-180.1°C (88% yield).

1H N.M.R. (300 MHz, DMSO) & ppm: 1.47 (d, 3H); 3.93 (q,
15 1H); 4.20 (s, 2H); 5.47 (s, 2H); 7.16 (d, 2H); 7.30 (dd,
1H); 7.42 (d, 1H); 7.49 (dd, 1H); 7.55 (d, 2H); 7.60
(dt, 1H); 7.75 (m, 4H); 7.83 (d, 1H); 7.97 (d, 1H); 8.03
(d, 1H); 8.40 (d, 1H); 10.11 (s, 1H).

Example 6: N-[4-(1H-5-tetrazolyl)methylphenyl]-2-[6
(2-quinolinylmethoxy)-2-naphthyllpropanamide (sodium salt)

1M sodium hydroxide (0.31 ml) was added to a solution of N-[4-(1H-5-tetrazolyl)methylphenyl]-2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamide (0.161 g, 0.31 mmol) in methanol (5 ml) and stirred at room temperature for 30 min. After that the reaction mixture was evaporated to dryness and the residue was recrystallized from ethanol-ethyl ether, to obtain 0.143 g of the title compound as a white solid with melting point 278-279°C (85% yield).

¹H N.M.R. (300 MHz, DMSO) 8 ppm: 1.47 (d, 3H); 3.93 (m,

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3H); 5.47 (s, 2H); 7.16 (d, 2H); 7.30 (dd, 1H); 7.42 (d, 1H); 7.49 (dd, 1H); 7.55 (d, 2H); 7.60 (dt, 1H); 7.75 (m, 4H); 7.83 (d, 1H); 7.97 (d, 1H); 8.03 (d, 1H); 8.40 (d, 1H); 10.11 (s, 1H).

5 Example 7: methyl 4-[2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propanamidolbenzoate

Following the process described in example 1 (point 2-[6-(2-quinolinylmethofrom starting F), xy)-2-naphthyl]propionic acid and methyl 4-aminobenzoate, the title compound was prepared as a white 10 solid with melting point 182.0-182.3°C (72% yield). 1 H N.M.R. (300 MHz, CDCl₃) δ ppm: 1.58 (dd, 3H); 3.81 (m, 4H); 5.43 (s, 2H); 7.18 (s, 1H); 7.24 (m, 1H); 7.35 (dd, 1H); 7.50 (m, 3H); 7.75 (m, 8H); 8.10 (d, 1H); 8.17 15 (d, 1H); 10.41 (s, 1H).

Example 8: 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyll-propanamido]benzoic acid

Following the process described in example 1 (point C), starting from methyl 4-[2-[6-(2-quinolinyl-methoxy)-2-naphthyl]propanamido]benzoate, the title compound was prepared as a white solid (recrystallized from ethanol) which decomposes at 264-265°C (74% yield).

1H N.M.R. (300 MHz, DMSO) 5 ppm: 1.49 (d, 3H); 4.00 (q, 1H); 5.48 (s, 2H); 7.32 (dd, 1H); 7.42 (d, 1H); 7.58 (d, 1H); 7.75 (m, 10H); 7.99 (d, 1H); 8.04 (d, 1H); 8.42 (d, 1H); 10.41 (s, 1H).

Example 9: methyl 4-[4-[2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propanamidolphenyl]butanoate

9A Methyl 4-(4-aminophenyl)butanoate

30 8.4 ml of concentrated H₂SO₄ was added to a solution of 4-(4-aminophenyl)butanoic acid (2.0 g, 11.6

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mmo) in methanol (84 ml) and refluxed for 2 h. After that the mixture was left to cool at room temperature, added with Na₂CO₃ to basic pH and extracted with ethyl acetate. The organic phase was dried and the solvent was evaporated off, to obtain 1.7 g of the title compound as a colourless oil (82% yield).

1H N.M.R. (300 MHz, CD₃OD) & ppm: 1.83 (q, 2H); 2.28 (t, 2H); 2.49 (t, 2H); 3.62 (s, 3H); 6.66 (d, 2H); 6.92 (d, 2H).

9B Methyl 4-[4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidolphenyl]butanoate

Following the process described in example 1 (point F), starting from 2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and methyl 4-(4-amino-phenyl)butanoate, the title compound was prepared as a white solid with melting point 145.7-149.0°C (87% yield).

1H N.M.R. (300 MHz, CDCl₃) 5 ppm: 1.61 (d, 3H); 1.82 (q, 2H); 2.27 (t, 2H); 2.56 (t, 2H); 3.62 (s, 3H); 3.81 (m, 1H); 5.42 (s, 2H); 7.02 (d, 2H); 7.28 (dd, 1H); 7.37 (m, 3H); 7.70 (m, 7H); 7.81 (d, 1H); 8.12 (d, 1H); 8.19 (d, 1H).

Example 10: 4-[4-[2-[6-(2-quinolinylmethoxy)-2-na-phthyl]propanamidolphenyl]butanoic acid

- Following the process described in example 1 (point C), starting from methyl 4-[4-[2-[6-(2-quino-linylmethoxy)-2-naphthyl]propanamido]phenyl]butanoate, the title compound was prepared as a white solid with melting point 176.0-176.3°C (68% yield).

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2H); 7.03 (d, 2H); 7.26 (dd, 1H); 7.38 (d, 1H); 7.48 (m, 3H); 7.57 (t, 1H); 7.72 (m, 5H); 7.95 (d, 1H); 8.00 (d, 1H); 8.38 (d, 1H).

Example 11: methyl 4-[2-[6-(2-quinolinylmetho-xy)-2-naphthyl]propanamidomethyl]benzoate

11A Methyl 4-cyanobenzoate

Following the process described in example 9 (point A), starting from 4-cyanobenzoic acid, the title compound was prepared as a yellowish oil (93% yield).

11B Methyl 4-aminomethylbenzoate

Following the process described in example 1 (point E), starting from methyl 4-cyanobenzoate and reacting for 4 h, the title compound was prepared as a semi-solid oil (87% yield).

¹H N.M.R. (300 MHz, CD₃OD) δ ppm: 3.91 (s, 3H); 4.21 (s, 2H); 7.59 (d, 2H); 8.07 (d, 2H).

11C Methyl 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyl]benzoate

Following the process described in example 1 (point F), starting from 2-(6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and methyl 4-aminomethyl-benzoate, the title compound was prepared as a white solid with melting point 149.0-150.8°C (68% yield).

¹H N.M.R. (300 MHz, CDCl₃) 8 ppm: 1.09 (d, 3H); 3.79 (s, 3H); 3.85 (q, 1H); 4.32 (s, 2H); 5.61 (s, 2H); 7.08 (m,

4H); 7.30 (m, 2H); 7.47 (m, 2H); 7.62 (m, 3H); 7.74 (t,

1H); 7.82 (d, 1H); 7.89 (d, 1H); 8.45 (d, 1H); 8.52 (d,

30 1H).

Example 12: 4-[2-[6-(2-Quinolinylmethoxy)-2-naphthyl]-

propanamidomethyllbenzoic acid

Pollowing the process described in example 1 (point C), starting from methyl 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyl]benzoate,

the title compound was prepared as a white solid with 5 melting point 203.9-205.8°C (68% yield).

¹H N.M.R. (300 MHz, CD_3OD) 8 ppm: 1.55 (d, 3H); 3.85 (q, 1H); 4.40 (q, 1H); 5.66 (s, 2H); 7.22 (d, 2H); 7.36 (dd, 1H); 7.41 (d, 1H); 7.47 (dd, 1H); 7.80 (m, 6H); 8.01 (m,

2H); 8.16 (d, 1H); 8.24 (d, 1H); 8.81 (d, 1H). Example 13: methyl 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyllpropanamidomethyllbenzoate

13A Methyl 3-cyanobenzoate

Following the process described in example 9 (point 15 A), starting from 3-cyanobenzoic acid, the compound was prepared as a yellowish oil (80% yield). ¹H N.M.R. (300 MHz, CDCl₃) δ ppm: 3.93 (s, 3H); 7.56 (t, 1H); 7.81 (dd, 1H); 8.23 (dd, 1H); 8.29 (d, 1H).

13B Methyl 3-aminomethylbenzoate

- 20 Following the process described in example 1 (point E), starting from methyl 3-cyanobenzoate and reacting for 4 h, the title compound was prepared as a semi-solid oil (92% yield).
- ¹H N.M.R. (300 MHz, CD₃OD) δ ppm: 3.88 (s, 3H); 3.91 (s, 2H); 7.49 (t, 1H); 7.51 (dd, 1H); 7.91 (dd, 1H); 7.99 25 (d, 1H).

13C Methyl 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyllbenzoate

Following the process described in example 1 (point 30 F). starting from 2-[6-(2-quinolinylmethoxy)-2-naphthyl]propionic acid and methyl 3-aminome-

thylbenzoate, the title compound was prepared as a white solid with melting point 135.4-136.8°C (63% yield).

¹H N.M.R. (300 MHz, CDCl₃) & ppm: 1.56 (d, 3H); 3.69 (q,

1H); 3.79 (s, 3H); 4.36 (d, 2H); 5.43 (s, 2H); 7.16 (d,

5 1H); 7.27 (m, 4H); 7.50 (t, 1H); 7.70 (m, 8H); 8.05 (d, 1H); 8.13 (d, 1H).

Example 14: 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyl]benzoic acid

Following the process described in example 1 (point 10 C), starting from methyl 3-[2-[6-(2-quino-linylmethoxy)-2-naphthyl]propanamidomethyl]benzoate, the title compound was prepared as a white solid with melting point 203.5-204.7°C (77% yield).

¹H N.M.R. (300 MHz, CDCl₃) 5 ppm: 1.43 (d, 3H); 3.80 (q,

15 1H); 4.27 (q, 2H); 7.25 (m, 3H); 7.42 (m, 2H); 7.62 (m, 1H); 7.75 (m, 5H); 8.00 (d, 1H); 8.05 (d, 1H); 8.32 (s, 1H); 8.42 (d, 1H); 8.56 (t, 1H).

Example 15: methyl 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyllpropanamidolphenylacetate

20 15A Methyl 4-aminophenylacetate

Following the process described in example 9 (point A), starting from 4-aminophenylacetic acid, the title compound was prepared as a yellowish oil (73% yield).

¹H N.M.R. (300 MHz, CDCl₃) 8 ppm: 3.50 (s, 2H); 3.65 (s, 3H); 3.68 (s, 2H); 6.59 (d, 2H); 7.04 (d, 2H).

15B Methyl 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]-

propanamidolphenylacetate

Following the process described in example 1 (point F), starting from 2-(6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and methyl 4-amino-phenylacetate, the title compound was prepared as a

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white solid with melting point 154.0-156.0°C (80% yield).

1_{H N.M.R.} (300 MHz, CDCl₃) 8 ppm: 1.64 (d, 3H); 3.53 (s, 2H); 3.64 (s, 3H); 3.83 (q, 1H); 5.50 (s, 2H); 7.03 (m,

5 1H); 7.15 (d, 2H); 7.31 (m, 4H); 7.56 (t, 1H); 7.72 (m,

5H); 7.83 (d, 1H); 8.11 (d, 1H); 8.19 (d, 1H).

Example 16: 4-[2-[6-(2-quinolinylmethoxyl-2-naphthyll-propanamidolphenylacetic acid

Following the process described in example 1 (point 10 C), starting from methyl 4-[2-[6-(2-quinolinyl-methoxy)-2-naphthyl]propanamido]phenylacetate, the title compound was prepared as a white solid with melting point 141.0-143.2°C (77% yield).

1_{H N.M.R.} (300 MHz, CDCl₃-CD₃OD) δ ppm: 1.69 (d, 3H);
3.63 (s, 2H); 4.04 (q, 1H); 5.60 (s, 2H); 7.28 (d, 2H);

7.39 (d, 2H); 7.59 (d, 3H); 7.72-7.80 (m, 2H); 7.86-7.96 (m, 4H); 8.05 (d, 1H); 8.22 (d, 1H); 8.53 (d, 1H).

Example 17: methyl 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyllpropanamidolphenylacetate

20 17A Methyl 3-aminophenylacetate

Following the process described in example 9 (point A), starting from 3-aminophenylacetic acid, the title compound was prepared as a yellowish oil (81% yield).

¹H N.M.R. (300 MHz, CDCl₃) 6 ppm: 3.52 (s, 2H); 3.67 (s, 3H); 3.68 (s, 2H); 6.55 (m, 2H); 6.65 (d, 1H); 7.09 (t, 1H).

17B Methyl 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidolphenylacetate

Following the process described in example 1 (point 30 F), starting from 2-(6-(2-quinolinylmetho-xy)-2-naphthyl]propionic acid and methyl 3-amino-

phenylacetate, the title compound was prepared as a white solid with melting point 141.8-142.8°C (75% yield).

¹H N.M.R. (300 MHz, CDCl₃) δ ppm: 1.65 (d, 3H); 3.54 (s, 2H); 3.64 (s, 3H); 3.82 (q, 1H); 5.51 (s, 2H); 6.96 (d, 1H); 7.03 (s, 1H); 7.35 (m, 6H); 7.56 (t, 1H); 7.75 (m, 6H); 8.12 (d, 1H).

Example 18: 3-[2-[6-(2-Quinolinylmethoxy)-2-naphthyl]propanamidolphenylacetic acid

- 10 Following the process described in example 1 (point C), starting from methyl 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]phenylacetate, the title compound was prepared as a white solid with melting point 200.9-202.9°C (61% yield).
- 20 Example 19: methyl 4-[4-[2-[6-[(7-chloro-2-quinoli-nyl)methoxyl-2-naphthyl]propanamidolphenyl]butanoate

 19A 2-Bromomethyl-7-chloroquinoline

N-bromosuccinimide (3.3 g, 18.5 mmol) and some crystals of 2,2'-azobis(2-methylpropionitrile) (AIBN) 25 added were to a solution of 7-chloro-2-methylquinoline (3 g, 16.9 mmol) in dry carbon tetrachloride (100 ml). The reaction mixture was refluxed for 4 h, then cooled at room temperature. The precipitate was filtered off and the filtrate was washed 30 with a NaCl saturated solution (3x20 ml), dried and the solvent was evaporated off to obtain a crude which was

purified by flash chromatography through a silica gel column. Eluting with petroleum ether:chloroform, 3:2, 2.3 g of the title compound were obtained as a white solid with melting point 110-111°C (52% yield).

19B Methyl 2-[6-(7-chloro-2-quinolinyl)methoxy-2-naphthyllpropionate

- Pollowing the process described in example 1 (point B), starting from methyl 2-(6-hydroxy-2-na-phthyl)propionate and 2-bromomethyl-7-chloroquinoline, the title compound was prepared as a white solid with melting point 98-99°C (78% yield)

19C 2-[6-(7-Chloro-2-quinolinyl)methoxy-2-naphthyl]pro20 panoic acid

Following the process described in example 1 (point C), starting from methyl 2-[6-(7-chloro-2-quinolinyl)methoxy-2-naphthyl]propionate, the title compound was prepared (90% yield).

19D Methyl 4-[4-[2-[6-[(7-chloro-2-quinolinyl)metho-

30 <u>xyl-2-naphthyllpropanamidolphenyllbutanoate</u>

Pollowing the process described in example 1 (point

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F), starting from 2-[6-(7-chloro-2-quino-linyl)methoxy-2-naphthyl]propanoic acid and methyl 4-(4-aminophenyl)butanoate, the title compound was prepared as a white solid with melting point 131.0-133.0°C (60% yield).

¹H N.M.R. (300 MHz, CDCl₃) 8 ppm: 1.58 (d, 3H); 1.80 (q, 2H); 2.20 (t, 2H); 2.48 (t, 2H); 3.56 (s, 3H); 3.75 (q, 1H); 5.41 (s, 2H); 7.06 (m, 3H); 7.23 (dd, 1H); 7.30 (m, 3H); 7.41 (dd, 1H); 7.50 (dd, 1H); 7.73 (m, 4H); 8.10 (d, 1H); 8.16 (d, 1H).

Example 20: 4-[4-[2-[6-[(7-chloro-2-quinoliny1)-metho-xy]-2-naphthyl]propanamidolphenyl]butanoic acid

Following the process described in example 1 (point C), starting from methyl 4-[4-[2-[6-[(7-chloro-2-quinolinyl)methoxy]-2-naphthyl]propanamido]phenyl]butanoate, the title compound was prepared as a white solid with melting point 176.4-177.8°C (69% yield).

1H N.M.R. (300 MHz, DMSO) & ppm: 1.42 (d, 3H); 1.69 (q, 2H); 2.11 (t, 2H); 3.89 (q, 1H); 5.43 (s, 2H); 7.03 (d, 2H); 7.26 (dd, 1H); 7.37 (d, 1H); 7.45 (m, 3H); 7.61 (dd, 1H); 7.71 (m, 3H); 7.81 (d, 1H); 8.01 (d, 1H); 8.05

Example 21: N-[4-(1H-5-tetrazolv1)phenylmethy1]--6-(2-quinolinylmethoxy)-2-naphthaleneacetamide

25 <u>21A 2-Acetyl-6-methoxynaphthalene</u>

(s, 1H); 8.42 (d, 1H); 9.99 (s, 1H).

2-methoxynaphthalene (10 g, 63.3 mmol) followed by acetyl chloride (5.8 ml, 79.8 mmol) were added drop by drop to a solution of aluminium trichloride (10.9 g, 81.7 mmol) in nitrobenzene (50 ml) cooled at 0°C and under inert atmosphere. The reaction mixture was stirred at 0°C for 2 h and at room temperature for 18 h, then it

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was cooled to 0°C, poured onto ice (50 ml), added with concentrated HCl (20 ml) and chloroform (25 ml). The two phases were separated and the organic one was washed with water (3x10 ml), dried and the solvent was evaporated off, to obtain a crude which was purified by crystallization in methanol, thereby obtaining 7.9 g of the title compound as a white solid with melting point 107-109°C (63% yield).

 1 H N.M.R. (300 MHz, CDCl₃) & ppm: 2.69 (s, 3H); 3.93 (s, 10 3H); 7.17 (m, 2H); 7.75 (d, 1H); 7.83 (d, 1H); 7.99 (dd, 1H); 8.37 (s, 1H).

21B Methyl 6-methoxy-2-naphthaleneacetate

mixture consisting of 2-acety1-6-methoxynaphthalene (3 g, 15.0 mmol), sulfur (0.72 g, 22.5 mmol) and morpholine (2 ml) was refluxed for 18 h, then acetic acid (11 ml) and concentrated HCl (18 ml) were added and reflux was continued for a further 24 h. After that the mixture was evaporated to dryness, added with methanol (60 ml) and concentrated H2SO4 (10 ml) and refluxed for 18 h. Finally the mixture was evaporated to dryness, added with ethyl acetate, washed with a 5% NaHCO3 saturated solution until the washing were neutral, dried and the solvent was evaporated off, to obtain a crude, which was purified by flash chromatography through a 25 silica column. Eluting with gel petroleum ether:chloroform, 4:1, 2.5 g of the title compound were obtained as a white solid with melting point 75.7-76.5°C (72% yield).

 1 H N.M.R. (300 HHz, CDCl₃) & ppm: 3.90 (s, 3H); 3.96 (s, 30 2H); 4.11 (s, 3H); 7.34 (m, 2H); 7.57 (dd, 1H); 7.85 (s, 1H); 7.88 (d, 1H); 7.90 (d, 1H).

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21C Methyl 6-hydroxy-2-naphthaleneacetate

Following the process described in example 1 (point A), starting from methyl 6-methoxy-2-naphthaleneacetate, the title compound was prepared as a yellowish solid with melting point 84.8-85.8°C (72% yield)

1H N.M.R. (300 MHz, CDCl₃) 6 ppm: 3.74 (s, 3H); 3.77 (s, 2H); 6.16 (m, 1H); 7.03 (m, 2H); 7.32 (dd, 1H); 7.56 (m, 3H).

21D Methyl 6-(2-quinolinylmethoxy)-2-naphthaleneacetate

10 Following the process described in example 1 (point B), starting from methyl 6-hydroxy-2-naphthaleneacetate, the title compound was prepared as a white solid with melting point 106.3-107.9°C (82% yield)

¹H N.M.R. (300 MHz, CDCL₃) 8 ppm: 3.72 (s, 3H); 3.78 (s, 2H); 5.64 (s, 2H); 7.26 (d, 1H); 7.33 (dd, 1H); 7.38 (dd, 1H); 7.59 (t, 1H); 7.74 (m, 5H); 7.86 (d, 3H); 8.15 (d, 1H); 8.22 (d, 1H).

21E 6-(2-Quinolinylmethoxy)-2-naphthaleneacetic acid

Following the process described in example 1 (point 20 C), starting from methyl 6-(2-quinolinylme-thoxy)-2-naphthaleneacetate, the title compound was prepared (84% yield).

1H N.M.R. (300 MHz, DMSO) & ppm: 3.75 (s, 2H); 5.60 (s, 2H); 7.20 (d, 1H); 7.31 (dd, 1H); 7.35 (dd, 1H); 7.60 (t, 1H); 7.68-7.83 (m, 6H); 8.20 (d, 1H); 8.25 (d, 1H).

21F N-[4-(1H-5-Tetrazolyl)phenylmethyl]-6-(2-quinolinylmethoxy)-2-naphthaleneacetamide

Following the process described in example 1 (point F), starting from 6-(2-quinolinylmetho-xy)-2-naphthaleneacetic acid and 4-(1H-5-tetrazo-lyl)benzylamine hydrochloride, the title compound was

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prepared as a white solid with melting point 256-257°C (58% yield).

¹H N.M.R. (300 MHz, DMSO) & ppm: 3.66 (s, 2H); 4.39 (d, 2H); 5.31 (s, 2H); 7.40 (m, 5H); 7.67 (t, 1H); 7.82 (m, 6H); 7.99 (m, 3H); 8.47 (d, 1H); 8.73 (t, 1H).

Example 22; methyl 4-[4-[6-(2-quinolinylmethoxy)--2-naphthyllcarboxamidolphenyllbutanoate

22A 2-Cyano-6-hydroxynaphthalene

Following the process described in example 1 (point 10 A), starting from 2-cyano-6-methoxynaphthalene, the title compound was prepared (91% yield).

¹H N.M.R. (300 MHz, CD₃OD) δ ppm: 7.17 (m, 2H); 7.51 (dd, 1H); 7.74 (d, 1H); 7.82 (d, 1H); 8.22 (s, 1H).

22B 2-Cyano-6-(2-quinolinylmethoxy)naphthalene

Following the process described in example 1 (point B), starting from 2-cyano-6-hydroxynaphthalene, the title compound was prepared as a white solid with melting point 155.0-155.8°C (70% yield).

¹H N.M.R. (300 MHz, CDCl₃) 6 ppm: 5.30 (s, 2H); 7.34 (d, 1H); 7.46 (dd, 1H); 7.60 (m, 2H); 7.80 (m, 5H); 8.18 (m, 2H); 8.27 (d, 1H).

22C 6-(2-Quinolinylmethoxy)-2-naphthoic acid

A solution of 35% sodium hydroxide (35 ml) was added to a solution of 2-cyano-6-(2-quinolinylmethoxy)-naphthalene (0.779, 2.5 mmol) in ethanol (130 ml) and refluxed for 24 h. After that the reaction was cooled at room temperature, added 1M HCl to pH 5 and left to stand at 5-10°C for 24 h. After this time, the resulting precipitate was filtered, washed with cold ethanol and dried over phosphorous pentoxide, thereby obtaining 0.796 g of the title compound as a white solid with

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melting point 239-240°C (96% yield).

¹H N.M.R. (300 MHz, DMSO) 6 ppm: 5.53 (s, 1H); 7.39 (dd, 1H); 7.54 (s, 1H); 7.62 (t, 2H); 7.76 (m, 3H); 8.00 (m, 4H); 8.42 (d, 1H); 8.53 (s, 1H).

5 <u>22D Methyl 4-[4-[[6-(2-quinolinylmethoxy)-2-naph-thyl]carboxamidolphenyl]butanoate</u>

Following the process described in example 1 (point F), starting from 6-(2-quinolinylmethoxy)-2-naphthoic acid and methyl <math>4-(4-aminophenyl)butanoate, the title compound was prepared (70% yield).

1H N.M.R. (300 MHz, DMSO) 6 ppm: 1.75 (q, 2H); 2.17 (t,
2H); 2.53 (t, 2H); 3.60 (s, 3H); 5.50 (s, 2H); 7.13 (d,
1H); 7.38 (dd, 1H); 7.53 (d, 1H); 7.59 (dt, 1H); 7.73
(m, 4H); 7.86 (d, 1H); 7.95 (m, 4H); 8.40 (d, 1H); 8.47
(s, 1H); 10.25 (s, 1H).

Example 23: 4-[4-[[6-(2-quinolinvlmethoxy)-2-naphthyl]-carboxamidolphenyllbutanoic acid

Following the process described in example 1 (point C), starting from methyl 4-[4-[[6-(2-quino-linylmethoxy)-2-naphthyl]carboxamido]phenyl]butanoate, the title compound was prepared as a white solid with melting point 231.9-232.5°C (78% yield).

¹H N.M.R. (300 MHz, DMSO) 5 ppm: 1.75 (q, 2H); 2.17 (t, 2H); 2.53 (t, 2H); 5.50 (s, 2H); 7.13 (d, 1H); 7.38 (dd, 1H); 7.52 (d, 1H); 7.53 (d,

25 1H); 7.53 (d, 1H); 7.59 (dt, 1H); 7.73 (m, 4H); 7.86 (d, 1H); 7.95 (m, 4H); 8.40 (d, 1H); 8.47 (s, 1H); 10.25 (s, 1H).

Example 24: N-[3-(1H-5-tetrazolyl)phenylmethyll-6-(2-quinolinylmethoxy)-2-naphthalenecarboxamide

Following the process described in example 1 (point F), starting from 6-(2-quinolinylmethoxy)-2-naphthoic

acid and 3-(1H-5-tetrazolyl)benzylamine, the title compound was prepared as a white solid with melting point 234.5-235.0°C (68% yield).

1_H N.M.R. (300 MHz, DMSO) 5 ppm: 4.63 (d, 2H); 5.53 (s, 2H); 7.41 (dd, 1H); 7.56 (m, 3H); 7.63 (t, 1H); 7.75 (d, 1H); 7.81 (dt, 1H); 7.92 (m, 3H); 8.01 (m, 4H); 8.45 (d, 1H); 8.48 (s, 1H); 9.26 (t, 1H).

Example 25: methyl 4-[4-[[7-(2-quinolinylmethoxy)-2-na-phthyl]carboxamido]phenyl]butanoate

10 25A 2-t-Butyldimethylsilyloxy-7-hydroxynaphthalene

t-butyldimethylsilyl chloride (4.71 g, 31.2 mmol) was added to a solution of 2,7-dihydroxynaphthalene (5 g, 0.031 mmol) and imidazole (1.9 g, 0.028 mmol) in dry DMF (30 ml), cooled at 0°C and under inert atmosphere. The reaction mixture was stirred for 2.5 h. After that 15 ethyl ether (100 ml) was added to the reaction mixture, to give a precipitate which was filtered off. filtrate was washed with a NaCl saturated solution (3x20 ml), dried and the solvent was evaporated off, to obtain a crude which was purified by flash chromatography 20 through a silica gel column, eluting with petroleum ether: ethyl ether to obtain 6.0 g of the title compound as a white solid with melting point 104.8-105.8°C (70% yield).

25 ¹H N.M.R. (300 MHz, CDCl₃) 8 ppm: 0.30 (s, 6H); 1.07 (s, 9H); 5.86 (m, 1H); 7.00 (m, 4H); 7.69 (d, 1H); 7.70 (d, 1H).

25B: 2-t-Butyldimethylsilyloxy-7-trifluoromethylsulfoxynaphthalene

30 Pyridine (1.66 ml) and trifluoromethanesulfonic anhydride (4.32 g, 15.2 mmol) was added to a solution of

2-t-butyldimethylsilyloxy-7-hydroxynaphthalene (3.5 g, 12.7 mmol) in methylene chloride (15 ml), cooled at 0°C and under inert atmosphere. The reaction mixture was stirred at this temperature for 1 h, then diluted with ethyl ether (100 ml), washed in succession with 0.01M HCl, a NaHCO₃ saturated solution and a NaCl saturated solution, dried and the solvent was evaporated off, to obtain 5 g of the title compound as an orange oil (97% yield).

10 ¹H N.M.R. (300 MHz, CDCl₃) δ ppm: 7.24 (m, 3H); 7.67 (s, 1H); 7.81 (d, 1H); 7.87 (d, 1H).

25C Methyl 7-t-butyldimethylsilyloxy-2-naphthalenecarboxylate

23.3 mmol), dimethyl ml. Triethylamine (3.2 sulfoxide (31 ml) Pd(OAc)2 (0.069 g, 0.31 mmol) and 15 1,3-bis(diphenylphosphino)propane (0.128 g, 0.31 mmol) were added to a solution of 2-t-butyldimethylsilyloxy-7trifluoromethylsulfoxynaphthalene (3.6 g, 10.6 mmol) in absolute methanol (20 ml), . The mixture was subjected to a carbon monoxide stream for 4 min, heated to a 75°C 20 for 3 h under carbon monoxide atmosphere, cooled at room temperature, filtered through celite and methanol was evaporated off. The resulting solution was diluted with ethyl ether and washed in succession with water, 5% HCl, a 5% NaHCO2 solution and a NaCl saturated solution, 25 dried and the solvent was evaporated off, to obtain a crude which was purified by flash chromatography through a silica gel column. Eluting with petroleum ether: ethyl ether 98:2, 2.0 g of the title compound were prepared as 30 a yellowish oil (60% yield).

 1 H N.M.R. (300 MHz, CDCl₃) & ppm: 0.27 (s, 6H); 1.04 (s,

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9H); 3.95 (s, 3H); 7.20 (dd, 1H); 7.32 (d, 1H); 7.84 (m, 3H); 8.44 (s, 1H).

25D Methyl 7-hydroxy-2-naphthalenecarboxylate

 K_2CO_3 (1.5 g, 12.5 mmol) was added to a solution of 5 methyl 7-t-bútyldimethylsilyloxy-2-naphthalenecarboxylate (1.30 g, 4.17 mmol) in tetrahydrofuran (40 ml) and methanol (40 ml) and stirred at room temperature for 4.5 h under nitrogen atmosphere. A NH₄Cl saturated solution and ethyl ether were added thereto, the phases were separated and the aqueous one was extracted with ethyl ether. The combined organic extracts were washed with water, dried and the solvent was evaporated off, to obtain а crude which was purified by chromatography through a silica gel column. Bluting with petroleum ether:ethyl acetate, 4:1, 0.96 g of the title compound were obtained (98% yield).

¹H N.M.R. (300 MHz, $CD_{3}OD$) δ ppm: 3.96 (s, 3H); 7.22 (m, 2H); 7.80 (m, 3H); 8.39 (s, 1H).

25E Methyl 7-(2-quinolinylmethoxy)-2-naphthalenecarboxylate

Following the process described in example 1 (point B). starting from methyl 7-hydroxy-2naphthalenecarboxylate, the title compound was prepared (97% yield).

25 ¹H N.M.R. (300 MHz, $CDCl_3$) 6 ppm: 3.98 (s, 3H); 5.55 (s, 2H); 7.36 (d, 1H); 7.43 (dd, 1H); 7.59 (dt, 1H); 7.81 (m, 5H); 7.95 (dd, 1H); 8.16 (d, 1H); 8.23 (d, 1H); 8.47 (s, 1H).

25F 7-(2-Quinolinylmethoxy)-2-naphthoic acid

30 Following the process described in example 1 (point starting C), from methyl 7-(2-quinolinylmetho-

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xy)-2-naphthalenecarboxylate, the title compound was prepared as a white solid with melting point 227.8-228.8°C (83% yield).

1H N.M.R. (300 MHz, CD₃OD-CDCl₃) 6 ppm: 5.52 (s, 2H);
7.42 (m, 2H); 7.62 (t, 1H); 7.76-7.95 (complex signal,
6H); 8.11 (d, 1H); 8.33 (d, 1H); 8.98 (s, 1H).

25G Methyl 4-[4-[[7-(2-quinolinylmethoxy)-2-naph-thyl]carboxamidolphenyl]butanoate

Following the process described in example 1 (point 10 F), starting from 7-(2-quinolinylmethoxy)-2-naphthoic acid and methyl 4-(4-aminophenyl)butanoate, the title compound was prepared as a white solid with melting point 166.4-167.9°C (64% yield).

1H N.M.R. (300 MHz, CDCl₃) 8 ppm: 1.84 (q, 2H); 2.23 (t,
2H); 2.52 (t, 2H); 7.05 (d, 2H); 7.12 (d, 1H); 7.26 (dd,
1H); 7.60 (m, 8H); 8.06 (m, 4H).

Example 26: 4-[4-[[7-(2-quinolinylmethoxy)-2-naphthyl]-carboxamidolphenyl]butanoic acid

Following the process described in example 1 (point 20 C), starting from methyl 4-[4-[[7-(2-quinolinyl-methoxy)-2-naphthyl]carboxamido]phenyl]butanoate, the title compound was prepared as a white solid with melting point 300-302°C (80% yield).

1H N.M.R. (300 MHz, DMSO) & ppm: 1.72 (q, 2H); 1.92 (t,
25 2H); 2.50 (m, 2H); 5.54 (s, 2H); 7.15 (d, 2H); 7.45 (dd,
1H); 7.59 (d, 1H); 7.70 (m, 6H); 7.90 (m, 4H); 8.33 (s,
1H); 8.44 (m, 2H).

Example 27: N-[4-(1H-5-tetrazolv1)phenylmethyl]-7-(2-quinolinylmethoxy)-2-naphthalenecarboxamide

Following the process described in example 1 (point F), starting from 7-(2-quinolinylmethoxy)-2-naphthoic

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acid and 4-(1H-5-tetrazolyl)benzylamine hydrochloride, the title compound was prepared as a white solid with melting point 224.4-225.9°C (87% yield).

¹H N.M.R. (300 MHz, DMSO) 5 ppm: 4.58 (d, 2H); 5.53 (s, 2H); 7.44 (dd, 1H); 7.54 (m, 3H); 7.62 (t, 1H); 7.74 (d, 1H); 7.81 (m, 2H); 7.99 (m, 6H); 8.36 (s, 1H); 8.43 (d, 1H); 9.24 (t, 1H).

Example 28: ethyl 4-[2-[[7-(2-quinolinylmethoxy)--2-naphthyl]carboxamidolethyl]benzoate

10 28A Methyl 4-cyanomethylbenzoate

Methyl 4-chloromethylbenzoate (8 g, 43.3 mmol) dissolved in ethanol (6 ml) was added to a solution of sodium cyanide (2.5 g, 51.0 mmol) in water (3 ml) and left at 100°C for 3 h. The reaction mixture was cooled at room temperature, added with ethyl ether (30 ml) and a NaCl saturated solution (10 ml). The two phases were separated and the aqueous one was extracted with ethyl ether (3x25 ml). The ether extracts were dried and the solvent was evaporated off, to obtain a crude which was purified by flash chromatography through a silica gel column. Bluting with petroleum ether:ethyl acetate, 3:2, 6.1 g of the title compound were prepared as a yellowish oil (80% yield).

¹H N.M.R. (300 MHz, CDCl₃) 8 ppm: 3.85 (s, 2H); 3.95 (s, 2H); 7.44 (d, 2H); 8.07 (d, 2H).

28B Methyl 4-(2-aminoethyl)benzoate

Following the process described in example 1 (point E), starting from methyl 4-cyanomethylbenzoate, the title compound was prepared as a semi-solid oil (90% yield).

¹H N.M.R. (300 MHz, CD₃OD) 5 ppm: 3.07 (t, 2H); 3.32 (t,

2H); 3.92 (s, 3H); 7.44 (d, 2H); 8.03 (d, 2H).

28C Ethyl 4-[2-[[7-(2-quinolinylmethoxy)-2-naphthyl]-carboxamidolethyl]benzoate

Following the process described in example 1 (point 5), starting from 7-(2-quinolinylmethoxy)-2-naphthoic acid and ethyl 4-(2-aminoethyl)benzoate, the title compound was prepared as a white solid with melting point 212.2-213.4°C (83% yield).

1H N.M.R. (300 MHz, CD₃OD-CDCl₃) 6 ppm: 3.02 (t, 2H);
10 3.74 (m, 2H); 3.90 (s, 3H); 5.49 (s, 2H); 6.97 (t, 1H);
7.34 (m, 3H); 7.61 (m, 2H); 7.71-7.86 (complex signal, 5H); 7.98 (d, 2H); 8.00 (m, 2H); 8.23 (d, 1H).

Example 29: 4-[2-[[7-(2-quinolinylmethoxy)-2-naphthy1]-carboxamidolethyl]benzoic acid

- Pollowing the process described in example 1 (point C), starting from ethyl 4-[2-[[7-(2-quinolinyl-methoxy)-2-naphthyl]carboxamido]ethyl]benzoate, the title compound was prepared as a white solid with melting point 228.0-229.0°C (63% yield).
- 25 Example 30: N-[4-(1H-5-tetrazolvl)phenylethyl]-7-(2-quinolinylmethoxy)-2-naphthalenecarboxamide

 30A Methyl 4-[2-(t-butoxycarbonylamino)ethyl]benzoate

1M NaOH solution (15.4 ml) and di-t-butyl dicarbonate (2.27 g, 10.4 mmol) were added to a solution of methyl 4-(2-aminoethyl)benzoate (1.5 g, 6.96 mmol) in dioxane (30 ml) and water (15 ml) at 0°C. The reaction

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mixture was stirred at room temperature for 18 h, keeping pH at 9-10 throughout the reaction by means of several additions of 1M NaOH. Dioxane was evaporated off and the aqueous residue was acidified to pH 3 with 1M HCl, extracted with ethyl acetate (3x50 ml), dried and the solvent was evaporated off to obtain 1.6 g of the title compound (92% yield).

¹H N.M.R. (300 MHz, CDCl₃) δ ppm: 1.42 (s, 9H); 2.83 (t, 2H); 3.38 (q, 2H); 3.88 (s, 3H); 7.27 (d, 2H); 7.98 (d, 2H).

30B 4-[2-(t-Butoxycarbonylamino)ethyl]benzoic acid

1M potassium hydroxide (34.1 ml) was added to a solution of methyl 4-[2-(t-butoxycarbonylami-no)ethyl]benzoate (1.7 g, 6.82 mmol) in ethanol (380 ml) and stirred under reflux for 30 min. After that ethanol was removed, and the resulting solid residue was redissolved in water (35 ml), adjusting to pH 4-5 with 10% acetic acid, thereby obtaining a precipitate which was separated by filtration, washed with ethyl ether and dried over phosphorous pentoxide, to obtain 1.3 g of the title compound (91% yield).

¹H N.M.R. (300 MHz, CD₃OD) δ ppm: 1.41 (s, 9H); 2.82 (t, 2H); 3.30 (m, 2H); 7.32 (d, 2H); 7.92 (d, 2H).

30C 4-[2-(t-Butoxycarbonylamino)ethyl]benzamide

Triethylamine (1.7 ml, 12.34 mmol) and ethyl chloroformate (0.64 ml, 6.79 mmol) were added to a solution of 4-[2-(t-butoxycarbonylaminoethyl]benzoic acid (1.45 g, 6.17 mmol) in dry THF (100 ml). The reaction mixture was stirred at room temperature for 30 min, then subjected to an ammonia stream for 30 min, evaporated to dryness and treated with chloroform to

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remove the ethyl chloroformate excess, thus obtaining 1.2 of the title compound (83% yield).

¹H N.M.R. (300 MHz, CD₃OD) δ ppm: 1.40 (s, 9H); 2.81 (t, 2H); 3.32 (m, 2H); 7.32 (d, 2H); 7.92 (d, 2H).

5 30D 4-[2-(t-Butoxycarbonylamino)ethyl]benzonitrile

a solution of 4-[2-(t-butoxy-carbonylamino)ethyl]benzamide (0.5 g, 2.13 mmol) in 2 ml of DMF was added to a solution of phosphorous oxychloride (1 ml, 2.13 mmol) in dry DMF (16 ml), kept at 0°C and under inert atmosphere for 30 min. The reaction mixture was stirred at room temperature for 24 h, then poured onto ice and extracted with ethyl acetate (4x25 ml), dried and the solvent was evaporated off, to obtain 0.35 g of the title compound (77% yield).

30E 5-[4-(2-t-Butoxycarbonylamino)phenyl]-1H-tetrazol

Following the process described in example 1 (point 20 D), starting from 4-[2-(t-butoxycarbonylamino)-ethyl]benzonitrile, the title compound was prepared (72% yield).

¹H N.H.R. (300 HHz, CD₃OD) 6 ppm: 1.37 (s, 9H); 3.30 (t, 2H); 3.50 (t, 2H); 7.39 (d, 2H); 7.88 (d, 2H).

25 30F 2-[4-(1H-5-Tetrazolyl)phenyl]ethylamine

Trifluoroacetic acid (0.37 ml, 4.8 mmol) was added to a solution of 5-[4-(2-t-butoxycarbonylamino)-phenyl]-1H-tetrazole (0.312 g, 1.20 mmol) in dry methylene chloride (4 ml) and stirred at room temperature for 18 h. After that the reaction mixture was evaporated to dryness to obtain 0.358 g of the title

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compound as the trifluoroacetic acid salt (99% yield). 1 H N.M.R. (300 MHz, CD₃OD) 8 ppm: 3.09 (t, 2H); 3.28 (t, 2H); 7.45 (d, 2H); 7.94 (d, 2H).

30G N-[4-(1H-5-Tetrazolyl)phenylethyl]-7-(2-quinolinyl-methoxy)-2-naphthalenecarboxamide

Following the process described in example 1 (point F), starting from 7-(2-quinolinylmethoxy)-2-naphthoic acid and 2-[4-(1H-5-tetrazolyl)phenyl]ethylamine, the title compound was prepared as a white solid with melting point 187.4-189.1°C (67% yield).

1_{H N.M.R.} (300 MHz, DMSO) & ppm: 2.97 (t, 2H); 3.55 (m, 2H); 5.55 (s, 2H); 7.41-7.51 (complex signal, 4H); 7.63 (t, 1H); 7.78 (m, 3H); 7.89-8.08 (complex signal, 6H); 8.27 (s, 1H); 8.43 (s, 1H); 8.72 (t, 1H).

15 Example 31: ethyl 4-[4-[6-(2-quinolinylmethoxy)-2-naphthyllmethylaminocarbonyllphenyllbutanoate

31A 2-Aminomethyl-6-methoxynaphthalene

Following the process described in example 1 (point E), starting from 2-cyano-6-methoxynaphthalene, the title compound was prepared (90% yield).

1H N.M.R. (300 MHz, CD₃OD) & ppm: 3.92 (s, 3H); 4.25 (s,
2H); 7.19 (dd, 1H); 7.28 (d, 1H); 7.50 (dd, 1H); 7.797.88 (m, 3H).

31B 2-Acetylaminomethyl-6-methoxynaphthalene

25 Triethylamine (2.9 ml, 20.5 mmol) and acetic anhydride (0.73 ml, 7.69 mmol) were added to a solution of 2-aminomethyl-6-methoxynaphthalene (1.23 g, 6.40 mmol) in chloroform (200 ml), cooled at -30°C. The reaction mixture was left at this temperature for 2 h, then cooled at room temperature and added with water (50 ml), the two phases were separated and the organic one

was washed with a 0.2M HCl solution, dried and the solvent was evaporated off, to obtain the title compound as a yellowish solid with melting point 163-165°C (86% yield).

31C 6-Acetylaminomethyl-2-naphthol

Pollowing the process described in example 1 (point 10 A), starting from 2-acetylaminomethyl-6-metho-xynaphthalene, the title compound was prepared as a white solid with melting point 219-222°C (93% yield).

1H N.M.R. (300 MHz, CD₃OD) & ppm: 2.09 (s, 3H); 4.50 (s, 2H); 7.06 (m, 2H); 7.32 (dd, 1H); 7.65 (m, 3H).

15 <u>31D 2-Acetylaminomethyl-6-(2-guinolinylmethoxy)naphtha-</u> lene

Following the process described in example 1 (point B), starting from 6-acetylaminomethyl-2-naphthol, the title compound was prepared as a semi-solid oil (65% yield).

¹H N.M.R. (300 MHz, CDCl₃) δ ppm: 1.93 (s, 3H); 4.49 (d, 2H); 5.44 (s, 2H); 5.85 (s, 1H); 7.25 (m, 3H); 7.51 (t, 1H); 7.67 (m, 5H); 7.78 (d, 1H); 8.07 (d, 1H); 8.15 (d, 1H).

25 31E 2-Aminomethyl-6-(2-cuinolinylmethoxy)naphthalene

6M HCl (2 ml) was added to a solution of 2-acetylaminomethyl-6-(2-quinolinylmethoxy)naphthalene (0.100 g, 0.281 mmol) in dioxane (10 ml) and refluxed for 18 h. After that the reaction mixture was cooled at room temperature, diluted with water (10 ml), 1M NaOH was added to basic pH and extracted with ethyl acetate.

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The organic extracts were dried and the solvent was evaporated off, to obtain 0.064 g of the title compound (73% yield).

¹H N.M.R. (300 MHz, CD₃OD) δ ppm: 3.89 (s, 2H); 5.34 (s, 2H); 7.25 (m, 2H); 7.36 (dd, 1H); 7.54 (t, 1H); 7.70 (m, 5H); 7.84 (d, 1H); 8.02 (d, 1H); 8.24 (d, 1H).

31F Bthvl 4-(4-formylphenyl)butanoate

Hexamethylenetetramine (1.6 g, 11.5 mmol) was added to a solution of ethyl 4-phenylbutanoate (2 g, 10.4 mmol) in trifluoroacetic acid (10 ml) and left at 80°C for 18 h. After that the reaction mixture was evaporated to dryness, added with a NaHCO₃ saturated solution (40 ml) and extracted with ethyl ether (4x50 ml). The ether extracts were dried and the solvent was evaporated off, to obtain a crude which was purified by flash chromatography through a silica gel column. Eluting with hexane:ethyl acetate, 9:1, 1.3 g of the title compound was prepared as a colourless oil (57% yield).

1H N.M.R. (300 MHz, CDCl₃) & ppm: 1.23 (t, 3H); 1.96 (m,
20 2H); 2.31 (t, 2H); 2.71 (t, 2H); 4.11 (q, 2H); 7.32 (d,
2H); 7.78 (d, 2H); 9.95 (s, 1H).

31G 4-(3-Ethoxycarbonylpropyl)benzoic acid

1 ml of Jones's reagent was added at 0°C to a solution of ethyl 4-(4-formylphenyl)butanoate (1.16 g, 5.29 mmol) in acetone (7 ml). The reaction mixture was stirred at room temperature for 18 h, then added with isopropanol (1 ml) and extracted with ethyl ether. The organic phase was dried and the solvent was evaporated off, to obtain 1.05 g of the title compound as a colourless oil (84% yield).

¹H N.M.R. (300 MHz, CDCl₃) 8 ppm: 1.09 (t, 3H); 1.81 (m,

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2H); 2.17 (t, 2H); 2.55 (t, 2H); 3.97 (q, 2H); 7.11 (d, 2H); 7.87 (d, 2H); 9.11 (m, 1H).

31H Ethyl 4-[4-[[6-(2-quinolinylmethoxy)-2-naphthyl]-methylaminocarbonyl]phenyl]butanoate

5 Following the process described in example 1 (point F), starting from 2-aminomethyl-6-(2-quinolinyl-methoxy)naphthalene and 4-(3-ethoxycarbonylpropyl)ben-zoic acid, the title compound was prepared (61% yield).

1H N.M.R. (300 MHz, CDCl₃) & ppm: 1.26 (t, 3H); 1.95 (m, 2H); 2.30 (t, 2H); 2.65 (t, 2H); 4.10 (q, 2H); 4.70 (d, 2H); 5.40 (s, 2H); 7.25 (m, 5H); 7.70 (m, 9H); 8.15 (t, 2H).

Example 32: 4-[4-[[6-(2-quinolinylmethoxy)-2-naphthyl]-methylaminocarbonyllphenyllbutanoic acid

- Following the process described in example 1 (point C), starting from ethyl 4-[4-[[6-(2-quinolinyl-methoxy)-2-naphthyl]methylaminocarbonyl]phenyl]butanoate, the title compound was prepared as a white solid with melting point 185.9-187.8 (80% yield).

Example 33: N-[4-(1H-5-tetrazolyl)phenylpropyll--7-(2-quinolinylmethoxy)-2-naphthalenecarboxamide

33A 3-(4-Bromophenyl)propan-1-ol

A solution of 4-bromocinnamic acid (5.0 g, 22 mmol) in 20 ml of dry ethyl ether was added to a suspension of aluminium lithium hydride (2.49 g, 66 mmol) in dry ethyl ether (130 ml) under inert atmosphere. The reaction mixture was stirred at room temperature for 2 hours,

then a NaCl saturated solution in water (80 ml) was slowly added, the two phases were separated and the aqueous one was extracted with ethyl acetate (3x50 ml). The organic extracts were dried and the solvent was evaporated off to obtain 3.60 g of the title compound as a yellowish oil (76% yield).

 1 H N.M.R. (300 MHz, CDCl₃) 8 ppm: 1.85 (m, 2H); 2.66 (t, 2H); 3.65 (t, 2H); 7.06 (d, 2H); 7.39 (d, 2H).

33B 3-(4-Cyanophenyl)propan-1-ol

A mixture of 3-(4-bromophenyl)propan-1-ol (2.0 g, 9.3 mmol), copper (I) cyanide (1.49 g, 16.7 mmol) and N-methylpyrrolidinone (13 ml) was stirred at 200°C for 2.5 hours. After that the reaction mixture was cooled at room temperature, poured onto a solution of diethylamine (30 g) and water (80 ml) and extracted with ethyl acetate (3x40 ml). The combined organic phases were dried and volatiles were removed, to obtain an oil from which N-methylpyrrolidinone was removed by distillation under high vacuum (0.5 torr, 85°C), thereby obtaining 0.78 g of the title compound (52% yield).

1H N.M.R. (300 MHz, CDCL3) 8 ppm: 1.85 (m, 2H); 2.42 (s

broad, 1H); 2.76 (t, 2H); 3.64 (t, 2H); 7.29 (d, 2H);

33C 3-(4-Cyanophenyl)propyl methanesulfonate

7.53 (d, 2H).

Triethylamine (0.54 ml, 4.03 mmol) and methanesulfonyl chloride (0.30 ml, 4.03 mmol) were added to a solution of 3-(4-cyanophenyl)propan-1-ol (0.50 g, 3.10 mmol) in dry methylene chloride (15 ml), cooled at 0°C and under inert atmosphere. The reaction mixture was stirred at 0°C for 2 hours, after that was diluted with methylene chloride (50 ml), washed in succession with

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0.05M HCl, with a NaCl saturated solution, dried and the solvent was evaporated off. 0.675 g of the title compound were obtained as a semi-solid oil (94% yield).

¹H N.M.R. (300 MHz, CDCl₃) & ppm: 1.89 (m, 2H); 2.63 (t, 2H); 2.85 (s, 3H); 4.05 (t, 2H); 7.14 (d, 2H); 7.38 (d, 2H).

33D 4-(1H-5-Tetrazolyl)azidopropylbenzene

Following the process described in example 1 (point D), starting from 3-(4-cyanophenyl)propyl methanesulfonate, the title compound was prepared (70% yield).

¹H N.M.R. (300 MHz, CDCl₃) δ ppm: 1.79 (m, 2H); 2.63 (t, 2H); 3.18 (t, 2H); 7.21 (d, 2H); 7.98 (d, 2H).

33E 3-[4-(1H-5-Tetrazolyl)phenyl]propylamine hydrochloride

Following the process described in example 1 (point E), starting from 4-(1H-5-tetrazolyl)azidopropylbenzene, the title compound was prepared as a white solid (crystallized from methanol) which decomposes at 251°C (87% yield).

¹H N.M.R. (300 MHz, CD₃OD) 6 ppm: 2.03 (m, 2H); 2.83 (t, 2H); 3.00 (t, 2H); 7.48 (d, 2H); 7.97 (d, 2H).

33F N-[4-(1H-5-Tetrazolyl)phenylpropyl]-7-(2-quinoli-nylmethoxy)-2-naphthalenecarboxamide

- Following the process described in example 1 (point F), starting from 7-(2-quinolinylmethoxy)-2-naphthoic acid and 3-[4-(1H-5-tetrazolyl)phenyl]propylamine, the title compound was prepared as a white solid with melting point 218.0-219.8°C (65% yield).
- 30 ¹H N.M.R. (300 MHz, CD₃OD) 8 ppm: 1.91 (m, 2H); 2.73 (t, 2H); 3.33 (t, 2H); 5.52 (s, 2H); 7.42 (dd, 1H), 7.45-

7.54 (complex signal, 3H); 7.63 (t, 1H); 7.72-7.84 (complex signal, 3H); 7.88-8.00 (complex signal, 4H); 8.03 (dd, 1H); 8.25 (d, 1H); 8.28 (s, 1H); 8.44 (d, 1H); 8.63 (t broad, 1H).

5 Biological activity tests

The antagonistic activity on LTD_4 of the compounds of the present invention is determined by means of an inhibition test of the $[^3H]-LTD_4$ receptor binding in guinea-pig lung membranes, and a test of inhibition of LTD_4 -induced contractions in the mienteric plexus of guinea-pig isolated ileum.

[3H]-LTD₄ receptor binding inhibition test in quinea-pig lung membranes

receptors, are purified following the method described by Mong et. al. (Mong et al., Prostaglandins, 28, 805 (1984)). These purified membranes (150 µg/ml) are added to an incubation mixture containing 10 mM of PIPES buffer (piperazin-N,N'-bis(2-ethanesulfonic acid) (pH 7.4), 10 mM of CaCl₂, 10 mM of 5 MgCl₂, 2 mM of cysteine, 2 mM of glycine, 0.5 nM of [³H]-LTD₄ (4700-6400 GBq/mmol) and different concentrations of the product under test in a final volume of 310 µl. The reaction mixture is incubated for 30 minutes at 25°C.

The radioligand bound to the membranes is separated from the free one by dilution with 4 ml washing buffer (10 mM Tris-HCl (pH 7.4) and 100 mM NaCl) at 0°C and filtration with Whatman GP/B filters, by means of a Brandel Cell Harvester. The filters are washed 4 times with a total volume of 16 ml of washing buffer at 0°C. The radioactivity present in the filters is determined

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by liquid scintillation.

The specific binding is defined as the difference between the total binding of $[^3H]-LTD_A$ and the nonspecific binding determined in the presence of 1 μM LTD4. The data obtained in the competition tests are analyzed by a computational program, which determines the inhibition constant of each compound (K;) by means of the Cheng-Prusoff equation (Cheng et al., Biochem. Pharmacol., 22, 3094 (1973)).

 $K_i = IC_{50} / (1 + [L] / K_d)$ 10

wherein IC_{50} is the concentration of compound which desplaces a 50% of the bound radioligand, [L] is the concentration of $[^3H]LTD_4$ free in the test and K_d is the dissociation constant of the LTD_4 obtained in an independent way by means of Scatchard analysis.

Table 1 shows some of the activity values found for the compounds of the present invention.

Inhibition test of the contractions induced by LTD4 in the mienteric plexus of guinea-pig isolated ileum.

The antagonistic activity of the compounds of the 20 present invention in the isolated organ was evaluated as its ability to inhibit the contraction caused by LTD_4 in the mienteric plexus of the ileum of Dunkin Hartley male albino guinea-pig, weighing 300-350 g (Cristol J.P. and Sirois P. Res. Commun. Chem. Pathol., 59, 423 (1988)).

The smooth muscle of guinea-pig ileum exhibits sensitivity to leukotrienes and especially to LTD4, which acts as primary mediator in the inflammatory and allergic response (Carnathan G.W. et al. Agents Actions, 20, 124 (1987)).

The mienteric plexus is extracted from a 2-3 cm

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segment of the terminal portion of the guinea-pig ileum, previously sacrificed by cervical dislocation. The plexus is put, at a tension of 0.5 g, in a 5 ml organ bath, containing a solution of Tyrode (137 mM NaCl, 2.7 mM KCl, 1.4 mM CaCl₂, 0.4 mM NaH₂PO₄, 11.9 mM NaHCO₃, 0.8 mM MgSO₄, 5.5 mM glucose), saturated with carbogen gas (95% O_2 -5% CO_2) at 37°C. The solution also contains indomethacin (3.3 μ M) and atropine (0.4 μ M) to remove the action of the intrinsic prostaglandins and the cholinergic responses.

After a 45 minute stabilization period a maximum isotonic response is obtained (100% contractile response) adding to the bath chamber the LTD_4 agonist (3 nM). This process is repeated until the same contraction response is obtained twice. The isometric measures are made in an isotonic transducer.

After stabilization is restored, the product under test is incubated at different concentrations (dissolved in 0.1% final concentration DMSO) for 2.5 minutes, and after that the contraction with LTD_4 is induced again.

The antagonistic activity is expressed as IC_{50} , the concentration of compound which reduces by 50% the maximum contraction.

Table 1 shows some of the values of activity found for the compounds of the present invention.

Table 1

		· -	
	Compound	Inhibition of the	Inhibition of
-	Example N°.	$[^3 ext{H}] ext{-LTD}_4$ receptor	contractions
		binding K _i (nM)	induced in the
5			ileum by LTD ₄
			IC ₅₀ (nM)
	1	9.2±2	13
	2	29.5±3	66
	, з	12.0±3	18
10	5	34.0±5	>100
	6	32.7±3	>100
	10	38.5±6.3	>100
	14	68.0±5	75
	18	134±18	100
15	20	190±21	100
	21	5.5±0.5	6
	23	30.0±1.2	>100
	24	28.0±3	100
	26	1.5±0.2	3 8
20	27	7.8±2	28
	29	1.0±0,1	8
	30	3.4±1.9	24
	32	44±15	100
	33	13.1±2	87

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CLAIMS

1. A compound of formula I,

$$R^{\frac{3}{4}}$$

$$N$$

$$0$$

$$1$$

$$R^{\frac{2}{4}}$$

$$0$$

$$1$$

$$R^{\frac{3}{4}}$$

$$R^{\frac{1}{2}}$$

$$R^{\frac{1}{$$

10 wherein:

the substituent containing A is bound to the 6- or 7position of the 2-naphthol system;

the substituent containing B is bound to the benzene ring at any free position;

- 15 -R¹ is hydrogen or methyl;
 - $-R^2$ is hydrogen, fluorine, chlorine or $-OCH_3$, which is bound to the naphthalene system at any positions except the 2- and the one occupied by the other substituent; $-R^3$ is hydrogen, fluorine, chlorine or bromine;
- 20 -A- is a -CO-NR⁴- or -NR⁴-CO- group, wherein R⁴ is hydrogen or methyl;

-B is a 5-tetrazolyl or $-\text{COOR}^5$ group, wherein R^5 is hydrogen, a (C_1-C_4) -alkyl or a phenylalkyl group of less than 10 carbon atoms;

25 m is 0 or 1;

n and p are integers from 0 to 6, with the proviso that n + p is less or equal to 6;

as well as the solvates and pharmaceutically acceptable salts thereof.

2. A compound according to claim 1, wherein R_2 is hydrogen, B is a 5-tetrazolyl or $COOR^5$ group, and R^5 is

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hydrogen, methyl, ethyl or benzyl.

- 3. A compound according to claim 1 or 2, wherein \mathbb{R}^3 is hydrogen or chlorine, and -A- is -CONH- or -NHCO-.
- 4. A compound according to any one of claims 1 to 3, wherein the substituent containing A is bound to the 6-position of the 2-naphthol system.
 - 5. A compound according to claim 4, wherein R^1 is hydrogen, m is 1, and -A- is -NHCO-.
- 6. A compound according to claim 4, wherein -A- is -CONH-
 - 7. A compound according to claim 6, wherein n and p are integers from 0 to 3.
 - 8. A compound according to any one of claims 1 to 3, wherein the substituent containing A is bound to the 7-position of the 2-naphthol system.
 - 9. A compound according to claim 8, wherein R^1 is hydrogen, m is 1 and -A— is -CONH—.
 - 10. A compound according to claim 8, wherein m is 0 and A is -CONH-.
- 20 11. A compound according to claim 10, wherein n and p are integers from 0 to 3.
 - 12. A compound according to claim 1 selected from the following ones:
 - N-[4-(1H-5-tetrazolyl)phenylmethyl]-2-[6-(2-quinolinyl-
- 25 methoxy)-2-naphthyl]propanamide;
 - N-[3-(1H-5-tetrazolyl)phenylmethyl]-2-[6-(2-quinolinyl-methoxy)-2-naphthyl]propanamide;
 - N-[2-(1H-5-tetrazolyl)phenylmethyl]-2-[6-(2-quinolinyl-methoxy)-2-naphthyl]propanamide;
- N-[4-(1H-5-tetrazolyl)methylphenyl]-2-[6-(2-quinolinyl-methoxy)-2-naphthyl]propanamide;

N-[4-(1H-5-tetrazolyl)methylphenyl]-2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamide (sodium salt); 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]benzoic acid; 5 4-[4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]phenyl]butanoic acid; 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidomethyl]benzoic acid: 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamidome-10 thyl]benzoic acid; 4-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]phenylacetic acid; 3-[2-[6-(2-quinolinylmethoxy)-2-naphthyl]propanamido]phenylacetic acid; 15 4-[4-[2-[6-[(7-chloro-2-quinolinyl)methoxy]-2-naphthyl]propanamido]phenyl]butanoic acid; N-[4-(1H-5-tetrazolyl)phenylmethyl]-6-(2-quinolinylmethoxy)-2-naphthaleneacetamide; 4-[4-[[6-(2-quinolinylmethoxy)-2-naphthyl]carboxamido]-20 phenyl]butanoic acid: N-[3-(1H-5-tetrazolyl)phenylmethyl]-6-(2-quinolinylmethoxy)-2-naphthalenecarboxamide; 4-[4-[[7-(2-quinolinylmethoxy)-2-naphthyl]carboxamido]phenyl]butanoic acid; 25 N-[4-(1H-5-tetrazolyl)phenylmethyl]-7-(2-quinolinylmethoxy > -2-naphthalenecarboxamide; 4-[2-[[7-(2-quinolinylmethoxy)-2-naphthyl]carboxamido]ethyl]benzoic acid; N-[4-(1H-5-tetrazolyl)phenylethyl]-7-(2-quinolinylme-30 thoxy)-2-naphthalenecarboxamide; 4-[4-[[6-(2-quinolinylmethoxy)-2-naphthyl]methylamino-

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carbonyl]phenyl]butanoic acid;

N-[4-(1H-5-tetrazolyl)phenylpropyl]-7-(2-quinolinylme-thoxy)-2-naphthalenecarboxamide.

13. A process for the preparation of the compounds of general formula I of claim 1, and of the pharmaceutically acceptable salts thereof, wherein:

a) when in formula I -A- is -CO-NR 4 -, then a compound of general formula II,

$$\begin{array}{c|c}
R^2 & R^1 \\
\hline
(CH)_{m} - COOH
\end{array}$$
II

wherein R^1 , R^2 , R^3 and m have the above defined meanings, is reacted with a compound III,

III

wherein R⁴, n and p have the above defined meanings and D can be equivalent to the group B in I or, when B in formula I is COOH, then D contains suitable carboxy-protecting group; the reaction between II and III being carried out in the presence of a carboxy-activating agent and a base, to obtain a compound of formula IVa,

$$\begin{array}{c|c}
R^{2} & R^{1} & 0 \\
R^{3} & R^{4} & CH_{2} & R^{4}
\end{array}$$

$$\begin{array}{c|c}
R^{2} & CH_{2} & R^{2} & CH_{2} & R^{4}
\end{array}$$

$$\begin{array}{c|c}
CH_{2} & R^{4} & CH_{2} & R^{4}
\end{array}$$

IVa

which is equivalent to I, or is converted into I by

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removing of the carboxy-protecting group;

b) when in formula I $-\lambda$ is $-NR^4$ -CO-, then a compound of general formula V,

$$\begin{array}{c|c}
R^2 & R^1 \\
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V

wherein R^1 , R^2 , R^3 , R^4 and m have the above defined meanings, is reacted with a compound VI

VI

wherein n, p and D have the above defined meanings; the reaction between V and VI being carried out in the presence of a carboxy-activating agent and a base, to obtain a compound of formula IVb.

 R^{2} $| CH_{2} - N - C - (CH_{2})_{n}$ $| CH_{2} - N - C - (CH_{2})_{n}$ | IVD

which coincides with I, or is converted into I as described above for the compound IVa;

- c) and, if desired, the compound of general formula I is converted into the desired salt, by treatment with a base or a suitable ion-exchanger, according to conventional methods.
- 30 14. The use of a compound of any one of claims 1 to 12 in the preparation of a medicament for the therapeutical

treatment of leukotriene-mediated diseases.

- 15. The use according to claim 14, wherein the leukotriene-mediated diseases are of inflammatory or allergic type.
- 5 16. The use according to claim 15, wherein the inflammatory or allergic diseases are: bronchial asthma, allergic rhinitis, allergic conjunctivitis, rheumatoid arthritis, osteoarthritis, tendinitis, bursitis or psoriasis.
- 10 17. The use according to claim 14, wherein the leukotriene-mediated diseases are of cardiovascular type.
 - 18. The use according to claim 17, wherein the diseases of cardiovascular type are: cardiac ischemia, cardiac
- infarction, coronary spasm, cardiac anaphylaxis, cerebral oedema or endotoxic shock.

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A. CLASS IPC 6	C07D401/1 A61K31/47 C07D215/	714 C07D21 3						
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"A" document defining the general state of the art which is not considered to be of particular relevance or the considered to be of particular relevance or the considered to be of particular relevance investion								
Sting		"X" document of particular relevance; the cannot be considered novel or cannot	be considered to					
which citatio	ent which may throw doubts on priority district) or is cited to exhibiting the publication date of another me or other special reason (as specified) and referring to an oral disciprare, use, exhibition or	involve an inventive step when the de "Y" document of particular relevance; the cannot be considered to involve an in- document is combined with one or or	claimed invention ventive step when the ore other such docu-					
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Date of the	ectual completion of the international search	Date of mailing of the international a	earch report					
1	6 November 1995	22. 11. 95						
Name and	mailing address of the ISA European Patent Office, P.B. 3818 Patentian 2 NL - 2280 HV Rijewijk	Authorized officer						

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